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# Ifma Study Guide

**J Spring**



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web when the reactants have more potential energy than the products have and the extra energy is released to the surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x_0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a a glider

between springs on an air track is an example of a horizontal mass spring system

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image text potential energy diagram answer the questions using the graph

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web answer see below explanation the rate of a chemical reaction  $a + b \rightarrow c + d$  is given in first approximation by  $v = k [a]^m [b]^n$  where  $k = A e^{-E_a/RT}$  where  $k$  depends by many factors as solvent geometry type of reaction  $E_a$  is the activation energy  $R$  is the universal constant of gases  $T$  is the absolute temperature

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