

Mixed Stoichiometry Practice

Name _____

Date _____ Period _____

Write and/or balance the following equations (remember the diatomic elements and to cross-check charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your answer to sig figs.

1. Potassium chlorate decomposes into potassium chloride and oxygen gas.

Balanced Equation:

2. How many moles of oxygen are produced when 3.0 moles of potassium chlorate decompose completely?

3. Butane (C_4H_{10}) undergoes combustion.

Balanced Equation: _____ $C_4H_{10}(l)$ + _____ $O_2(g)$ \rightarrow _____ $CO_2(g)$ + _____ $H_2O(l)$

4. How many grams of CO_2 are produced when 60 g of O_2 are reacted with an excess of butane?

5. Water decomposes into hydrogen gas and oxygen gas by electrolysis.

Balanced Equation:

6. How many grams of hydrogen will be produced when 6.0 moles of oxygen are produced?

7. How many grams of water are required to produce 9.00 grams of hydrogen?

Mixed Stoichiometry Practice Answer Key

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**

Mixed Stoichiometry Practice Answer Key:

Embracing the Tune of Term: An Mental Symphony within **Mixed Stoichiometry Practice Answer Key**

In a world consumed by displays and the ceaseless chatter of instantaneous interaction, the melodic splendor and emotional symphony created by the published word frequently fade into the backdrop, eclipsed by the constant noise and disturbances that permeate our lives. Nevertheless, set within the pages of **Mixed Stoichiometry Practice Answer Key** a marvelous literary value overflowing with fresh emotions, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, this captivating masterpiece conducts viewers on a psychological journey, well unraveling the concealed melodies and profound influence resonating within each carefully crafted phrase. Within the depths with this touching review, we shall examine the book is key harmonies, analyze its enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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