

## ANSWER KEY

### Mixed Stoichiometry Problems



a). How many moles of  $\text{H}_2$  would be required to produce 5.0 moles of water?

5.0 moles water

b). What mass of  $\text{H}_2\text{O}$  is formed when  $\text{H}_2$  reacts with 384 g of  $\text{O}_2$ ?

432g  $\text{H}_2\text{O}$



a). Balance this equation

Look above.

b). What mass of  $\text{H}_2\text{SO}_4$  would be required to react with 0.75 mol of  $\text{NaOH}$ ?

36.8 = 37g  $\text{H}_2\text{SO}_4$

d). If 30.0 g of  $\text{H}_2\text{SO}_4$  reacts with sodium hydroxide, what mass of water is produced?

11.0g  $\text{NaOH}$



a). How many grams of  $\text{SnF}_2$  are produced from the reaction of 30.0 g of  $\text{HF}$ ? 117g  $\text{SnF}_2$

b) How many moles of  $\text{HF}$  will be needed to react with 48 grams of Tin, Sn?

0.81 moles  $\text{HF}$

4. In a spacecraft, the carbon dioxide exhaled by astronauts can be removed by its reaction with lithium hydroxide,  $\text{LiOH}$ , according to the following chemical equation.



If 20.0 moles of  $\text{CO}_2$  is exhaled, the average amount exhaled by a person each day, how much (in grams)  $\text{Li}_2\text{CO}_3(\text{s})$  is produced?

148g  $\text{Li}_2\text{CO}_3$



a). Balance the equation.

See above.

b). How many moles of sodium are required to produce 5.0 moles of  $\text{NaOH}$ , sodium hydroxide?

5.0 moles  $\text{Na}$

c). How many moles of  $\text{H}_2$  are produced in the reaction with 5.0 grams of water?

0.14 moles  $\text{H}_2$

# Mixed Stoichiometry Problems 2 And Answer

**KJ Lindholm-Leary**



## **Mixed Stoichiometry Problems 2 And Answer:**

## Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Mixed Stoichiometry Problems 2 And Answer**

In a world inundated with screens and the cacophony of fast interaction, the profound energy and psychological resonance of verbal art frequently diminish in to obscurity, eclipsed by the continuous barrage of sound and distractions. However, nestled within the lyrical pages of **Mixed Stoichiometry Problems 2 And Answer**, a captivating perform of literary beauty that pulses with fresh thoughts, lies an unique trip waiting to be embarked upon. Published with a virtuoso wordsmith, this enchanting opus books readers on an emotional odyssey, lightly revealing the latent possible and profound impact stuck within the intricate web of language. Within the heart-wrenching expanse of this evocative evaluation, we can embark upon an introspective exploration of the book is central subjects, dissect its fascinating publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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