

Chemistry – Unit 7 Review: Chemical Reactions

- ___ 1. Numbers placed in front of formulas in an equation are called _____
- ___ 2. New substances formed in a chemical reaction are called _____
- ___ 3. The total number of atoms represented by the formula $\text{Fe}_3(\text{PO}_4)_2$ is _____
- ___ 4. The symbol that indicates a substance dissolved in water is _____
- ___ 5. Two white powders are mixed in a flask. The bottom of the flask feels cooler. This an example of what kind of reaction? _____
- ___ 6. The charge on lead in PbSO_4 is _____



- _____ 7. How many more D atoms should be on the right side of the equation?
- _____ 8. What is the formula of the missing product molecule?
- _____ 9. What are particles on the left side of the arrow called?
10. Write the completed chemical equation for the reaction pictured.

Balance the following reactions.



Modeling Chemistry Unit Answers

Julia Schneider



Modeling Chemistry Unit Answers:

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