

Chapter 13 Test: Ions in Aqueous Solution & Colligative Properties

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- ___B___ 1. A solute whose water solution conducts electricity is called a(n)
a. nonconductor.
b. electrolyte.
c. nonelectrolyte.
d. aqueous solution.
- ___B___ 2. Which solution has the highest boiling point?
a. 0.50 *m* glucose in water
b. 0.50 *m* potassium iodide in water
c. 0.20 *m* glucose in water
d. 0.40 *m* calcium chloride in water
- ___C___ 3. When an ionic solid dissolves in water, which of the following does *not* occur?
a. The compound dissociates.
b. Hydration occurs.
c. The compound ionizes.
d. The solution conducts electrical current.
- ___A___ 4. Colligative properties depend on the
a. number of solute particles present.
b. size of solute particles present.
c. mass of solute particles present.
d. charge on solute particles present.
- ___C___ 5. Nonvolatile solutes
a. depress both the freezing point and the boiling point.
b. elevate both the freezing point and the boiling point.
c. depress the freezing point and elevate the boiling point.
d. elevate the freezing point and depress the boiling point.
- ___C___ 6. Compared with the freezing-point depression of a 0.01 *m* $C_6H_{12}O_6$ solution, the freezing-point depression of a 0.01 *m* NaCl solution is
a. almost half.
b. exactly the same.
c. almost twice as great.
d. almost three times as great

Holt Chemistry Ch 13 Test Solutions

Tzimopoulo



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Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters, 1902 **Elementary Modern Chemistry** Wilhelm Ostwald, Harry Wheeler Morse, 1909 *Books in Print Supplement*, 2002 *AP Chemistry Premium, 2022-2023: Comprehensive Review with 6 Practice Tests + an Online Timed Test Option* Neil D. Jespersen, Pamela Kerrigan, 2021-07-06

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2000-12 **The British Journal of Photography** William Crookes, T. A. Malone, George Shadbolt, John Traill Taylor, William Blanchard Bolton, Thomas Bedding, 1912

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1. Understanding the eBook Holt Chemistry Ch 13 Test Solutions
 - The Rise of Digital Reading Holt Chemistry Ch 13 Test Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Ch 13 Test Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Ch 13 Test Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Ch 13 Test Solutions
 - Personalized Recommendations
 - Holt Chemistry Ch 13 Test Solutions User Reviews and Ratings

- Holt Chemistry Ch 13 Test Solutions and Bestseller Lists
- 5. Accessing Holt Chemistry Ch 13 Test Solutions Free and Paid eBooks
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 - Holt Chemistry Ch 13 Test Solutions eBook Subscription Services
 - Holt Chemistry Ch 13 Test Solutions Budget-Friendly Options
- 6. Navigating Holt Chemistry Ch 13 Test Solutions eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Ch 13 Test Solutions Compatibility with Devices
 - Holt Chemistry Ch 13 Test Solutions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Ch 13 Test Solutions
 - Highlighting and Note-Taking Holt Chemistry Ch 13 Test Solutions
 - Interactive Elements Holt Chemistry Ch 13 Test Solutions
- 8. Staying Engaged with Holt Chemistry Ch 13 Test Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Ch 13 Test Solutions
- 9. Balancing eBooks and Physical Books Holt Chemistry Ch 13 Test Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Ch 13 Test Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Ch 13 Test Solutions
 - Setting Reading Goals Holt Chemistry Ch 13 Test Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Ch 13 Test Solutions
 - Fact-Checking eBook Content of Holt Chemistry Ch 13 Test Solutions
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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