

**Concept Review****Section: What Is Matter?**

1. **Classify** the following as a homogeneous, *O*, or a heterogeneous, *E*, mixture.

- |       |                                    |       |                           |
|-------|------------------------------------|-------|---------------------------|
| _____ | <b>a.</b> a pail of sand and water | _____ | <b>d.</b> banana split    |
| _____ | <b>b.</b> air                      | _____ | <b>e.</b> chocolate syrup |
| _____ | <b>c.</b> human blood              | _____ | <b>f.</b> sea water       |

2. **Compare and contrast** atoms and molecules.

\_\_\_\_\_

\_\_\_\_\_

3. **Explain** the difference between a pure substance and a homogeneous mixture.

\_\_\_\_\_

\_\_\_\_\_

4. **Classify** each of the following as an element or a compound.

- |       |                                |
|-------|--------------------------------|
| _____ | <b>a.</b> benzene, $C_6H_6$    |
| _____ | <b>b.</b> aluminum, Al         |
| _____ | <b>c.</b> aspirin, $C_9H_8O_4$ |
| _____ | <b>d.</b> titanium, Ti         |
| _____ | <b>e.</b> acetylene, $C_2H_2$  |
| _____ | <b>f.</b> zinc, Zn             |

5. **Explain** why elements and compounds are pure substances.

\_\_\_\_\_

\_\_\_\_\_

6. **Determine** which of the following are pure substances and which are mixtures.

- |       |  |
|-------|--|
| _____ | <b>a.</b> salt water                   |
| _____ | <b>b.</b> isopropyl alcohol, $C_3H_8O$ |
| _____ | <b>c.</b> mercury, Hg                  |
| _____ | <b>d.</b> ammonia, $NH_3$              |
| _____ | <b>e.</b> an egg yolk                  |
| _____ | <b>f.</b> honey                        |

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**Richard Bailey**



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