

NUCLEAR CHEMISTRY

BALANCING NUCLEAR REACTIONS WORKSHEET

Predict the missing product or reactant in the following nuclear reactions. Determine the type of nuclear reaction (α emission, β emission, γ emission, positron emission, artificial transmutation, fission, or fusion) described.

	Type of Nuclear Reaction
1.) ${}^{42}_{19}\text{K} \rightarrow {}^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	1.)
2.) ${}^{239}_{94}\text{Pu} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$	2.)
3.) ${}^{235}_{92}\text{U} \rightarrow \underline{\hspace{2cm}} + {}^{231}_{90}\text{Th}$	3.)
4.) ${}^1_1\text{H} + {}^2_1\text{H} \rightarrow \underline{\hspace{2cm}}$	4.)
5.) ${}^6_3\text{Li} + {}^1_0\text{n} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$	5.)
6.) ${}^{27}_{13}\text{Al} + {}^4_2\text{He} \rightarrow {}^{30}_{15}\text{P} + \underline{\hspace{2cm}}$	6.)
7.) ${}^9_4\text{Be} + {}^1_1\text{H} \rightarrow \underline{\hspace{2cm}} + {}^4_2\text{He}$	7.)
8.) ${}^{37}_{19}\text{K} \rightarrow {}^{0}_{-1}\text{e} + \underline{\hspace{2cm}}$	8.)
9.) $\underline{\hspace{2cm}} + {}^1_0\text{n} \rightarrow {}^{142}_{56}\text{Ba} + {}^{91}_{36}\text{Kr} + 3 {}^1_0\text{n}$	9.)
10.) ${}^{238}_{92}\text{U} + {}^4_2\text{He} \rightarrow \underline{\hspace{2cm}} + {}^1_0\text{n}$	10.)
11.) ${}^{14}_6\text{C} \rightarrow {}^{14}_7\text{N} + \underline{\hspace{2cm}}$	11.)
12.) ${}^{187}_{75}\text{Re} + \underline{\hspace{2cm}} \rightarrow {}^{188}_{75}\text{Re} + {}^1_1\text{H}$	12.)
13.) ${}^{22}_{11}\text{Na} + \underline{\hspace{2cm}} \rightarrow {}^{22}_{10}\text{Ne}$	13.) ---
14.) ${}^{218}_{84}\text{Po} \rightarrow \underline{\hspace{2cm}} + {}^4_2\text{He}$	14.)
15.) ${}^{253}_{99}\text{Es} + {}^4_2\text{He} \rightarrow {}^1_0\text{n} + \underline{\hspace{2cm}}$	15.)

Write the Balanced nuclear equations for the alpha decay of:

a) Plutonium-234

b) Strontium-90

Write the balanced nuclear equations for the alpha, beta and gamma decay of Radium-226

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Ralph Thomas Myers



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1. Understanding the eBook Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - The Rise of Digital Reading Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Personalized Recommendations
 - Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers User Reviews and Ratings
 - Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers and Bestseller Lists
5. Accessing Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers Free and Paid eBooks
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- Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers Budget-Friendly Options
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 - Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Highlighting and Note-Taking Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Interactive Elements Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
- 8. Staying Engaged with Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
- 9. Balancing eBooks and Physical Books Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Setting Reading Goals Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Fact-Checking eBook Content of Introduction To Nuclear Chemistry Nuclear Equations Worksheet Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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