

MULTIPLE CHOICE QUESTIONS

CHEMICAL BONDING AND MOLECULAR STRUCTURE

1. C-O bond length is minimum in

- (a) CO_2
- (b) CO_3^{2-}
- (c) HCOO^-
- (d) CO**

2. Molecules are held together in a crystal by

- (a) hydrogen bond
- (b) electrostatic attraction
- (c) Van der Waal's attraction**
- (d) dipole-dipole attraction

3. Sp^3d^2 hybridization is present in $[\text{Co}(\text{NH}_3)_6]^{3+}$, find its geometry

- (a) octahedral geometry**
- (b) square planar geometry
- (c) tetragonal geometry
- (d) tetrahedral geometry

4. Find the molecule with the maximum dipole moment

- (a) CH_4
- (b) NH_3**
- (c) CO_2
- (d) NF_3

Multiple Choice Questions On Bond Energy

James C. Hill



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