

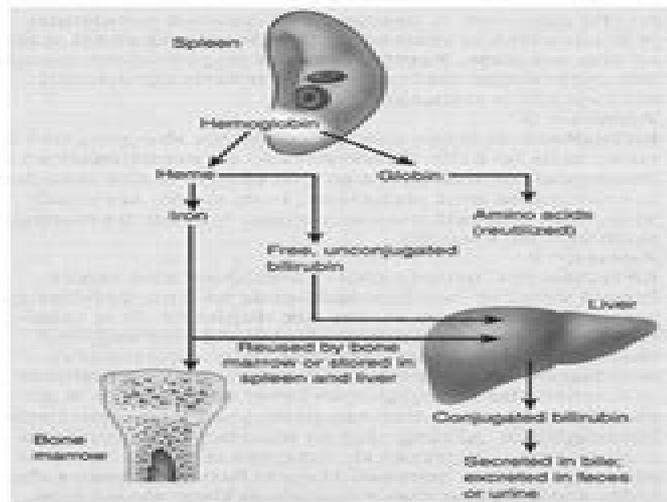
CHAPTER 13

SECTION I: ASSESSING YOUR UNDERSTANDING

Activity A

- biconcave, cell membrane
- iron
- nucleus
- 4
- glycolytic
- methemoglobin
- red blood cell count (RBC)
- hematocrit
- mean corpuscular hemoglobin concentration (MCHC)
- Anemia
- hypoxia
- Hemolytic
- sickle cell, thalassemias
- spherocytosis
- β -Thalassemias, α -thalassemias
- glucose-6-phosphatase (G6PD)
- iron deficiency
- chronic blood loss
- Megaloblastic
- Fernicious
- Aplastic anemia
- Polycythemia
- oxygen
- conjugate
- anemia

Activity B



Activity C

- e
- a
- f
- c
- i
- j
- b
- d
- h
- g

Activity D

- The hemoglobin molecule is composed of two pairs of structurally different α and β polypeptide chains. Each of the four-polypeptide chains consists of a globin (protein) portion and heme unit, which surrounds an atom of iron that binds oxygen. Thus, each molecule of hemoglobin can carry four molecules of oxygen. The binding that occurs is cooperative, or allosteric. When one oxygen molecule binds, it makes it easier for the next to bind. The process also works in reverse.
- A group of large phagocytic cells found in the spleen, liver, bone marrow, and lymph nodes facilitates the destruction of RBCs. These phagocytic cells recognize old and defective red cells and then ingest and destroy them in a series of enzymatic reactions. During these reactions, the amino acids from the globulin chains and iron from the heme units are salvaged and reused. The bulk of the heme unit is converted to bilirubin, which is insoluble in plasma and attaches to plasma proteins for transport. Bilirubin is removed from the blood by the liver and conjugated with glucuronide to render it water soluble so that it can be excreted in the bile.
- (1) Manifestations of impaired oxygen transport and the resulting compensatory mechanisms, (2) reduction in red cell indices and hemoglobin levels, and (3) signs and symptoms associated with the pathologic process that is causing the anemia
- Premature destruction of the cells due to the rigid nondeformable membrane occurs in the spleen, causing hemolysis and anemia due to a decrease in red cell numbers. Secondly, vessel occlusion, a complex process involving an interaction among the sickle cells, endothelial cells, leukocytes, platelets, and other plasma proteins, will interrupt blood flow. The adherence of sickle cells to the vessel endothelium causes endothelial activation with liberation of inflammatory mediators and substances that increase platelet activation and promote blood coagulation.
- Exposure to high doses of radiation, chemicals, and toxins that suppress cellular activity directly or through immune mechanisms is the standard cancer treatment. Chemotherapy and irradiation commonly result in bone marrow depression, which causes anemia, thrombocytopenia, and neutropenia. Identified toxic agents including benzene, the antibiotic chloramphenicol, and the alkylating agents and antimetabolites used in the treatment of cancer will decrease bone marrow of stem cells, thus affecting the production of RBCs.
- Viscosity rises exponentially with the hematocrit and interferes with cardiac output and blood flow. Hypertension is common, and there may

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Khayatpoor



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The Science Teacher ,1982 Some issues are accompanied by a CD ROM on a selected topic *Study Guide and Workbook, an Interactive Approach for Starr and McMillan's Human Biology, Third Edition* Jane B. Taylor,John D. Jackson,1999 Asks the student to write all answers in this study guide workbook This workbook is interactive because it requires students to do things instead of just read more material All questions are arranged by chapter modules so students may skip unassigned material Each module in the study guide refers to the page numbers of the corresponding module in the text There is a wide variety of questions multiple choice questions tables to be filled in art to be labeled true false questions requiring students to write the correct answer if the statement is false thought provoking conceptual questions boldfaced terms requiring a written definition list of objectives in fill in the blank format and other types of questions **Principles of Physical Chemistry** Hans Kuhn,Horst-Dieter Försterling,David H. Waldeck,2009-03-17 Principles of Physical Chemistry Second Edition uniquely uses simple physical models as well as rigorous treatments for understanding molecular and supramolecular systems and processes In this way the presentation assists students in developing an intuitive understanding of the subjects as well as skill in quantitative manipulations The unifying nature of physical chemistry is emphasized in the book by its organization beginning with atoms and molecules and proceeding to molecular assemblies of increasing complexity ending with the emergence of matter that carries information i e the origin of life a physicochemical process of unique importance The aim is to show the broad scope and coherence of physical chemistry **Core List of Books and Journals in Science and Technology** Russell H. Powell,James R. Powell,1987-10 *Study Guide for Man, Nature, and Society* Theodore W. Pohrte,L. Jack Pierce,1975

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