

Molarity Worksheet

1. Calculate the mass of $\text{CuSO}_4 \cdot 6\text{H}_2\text{O}$ required to prepare 200.0 mL of 0.300 M solution.
2. Calculate the mass of $\text{CoCl}_2 \cdot 8\text{H}_2\text{O}$ required to prepare 300.0 mL of a 0.520 M solution.
3. 150.0 g of NaCl is dissolved in 250.0 mL of water. Calculate the molarity.
4. 25.2 g of $\text{CuSO}_4 \cdot 6\text{H}_2\text{O}$ is dissolved in 28.0 mL of water. Calculate the molarity.
5. Calculate the mass of NaCl required to prepare 565.0 mL of a 0.450 M solution.
6. Calculate the volume of 0.250 M NaCl solution required to contain 0.0300 g of NaCl.
7. Calculate the volume of 0.500 M NaCl solution required to contain 0.52 g of NaCl.
8. Calculate the mass of NaCl required to prepare 360.0 mL of a 0.35 M solution.
9. 55.6 g of NaCl is dissolved in 562 mL of water. Calculate the molarity.
10. 78.9 g of $\text{CuSO}_4 \cdot 8\text{H}_2\text{O}$ is dissolved in 500.0 mL of water. Calculate the molarity.

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LL Leslie

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