

## THE MOLE: LAB ACTIVITY

### Station 1:

A) Determine the mass of one drop of water by measuring the mass of 50 drops of water.

Mass of dish (or beaker)	= _____ g
Mass of dish (or beaker) + 50 drops water	= _____ g
Mass of 50 drops water	= _____ g
Mass of 1 drop of water	= _____ g

B) How many molecules are there in one drop of water?

Remember 1 mole of a substance is  $6.02 \times 10^{23}$  particles (atoms or molecules).  
Therefore,  $6.02 \times 10^{23}$  molecules of water will weigh 18.02 g/mol

1. Find # of moles of water for 1 drop  $n = (\text{mass})/(\text{Molar Mass})$
2. # of molecules in one drop of water  $= n_{\text{water}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}}$   
 $= ? \text{ molecules}$

### Station 2

Calculate the number of copper atoms in a penny.

1. Find # of moles of Cu  $n = (\text{mass})/(\text{Molar Mass})$
2. # of Cu atoms in a penny  $= n_{\text{Cu}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}}$   
 $= ? \text{ atoms}$

### Station 3

A. Calculate the number of moles in one spoon of sugar. Assume that sugar is made up entirely of sucrose molecules ( $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ ).

$$M_{\text{sucrose}} = 342.34 \text{ g/mol}$$

$$n_{\text{sugar}} = \frac{m_{\text{sucrose}}}{M_{\text{sucrose}}}$$
$$= \underline{\hspace{2cm}} \text{ mol}$$

B. Calculate the number of molecules of sugar in one spoon of sugar.

$$\begin{aligned} \# \text{ of molecules sucrose} &= n_{\text{sucrose}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \\ &= \underline{\hspace{2cm}} \text{ mol} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}} \\ &= ? \text{ molecules} \end{aligned}$$

# Mole Lab Activity Answers

**McGraw-Hill Staff**



## **Mole Lab Activity Answers:**

**CBSE Chemistry Chapterwise Case Study Class 11** Priti Singhal,2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging *ChemDiscovery Lab Manual* Olga I. Agapova,2002

**Regents Exams and Answers: Chemistry--Physical Setting Revised Edition** Barron's Educational Series,Albert Tarendash,2021-01-05 Barron s Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day *Stoichiometry Unit Project* Luann Marie Decker,1998

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**Physics Guide and Lab Activities** Ira Cleveland Davis,1957

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Curriculum Development Library,1980

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Improving Student Comprehension in Chemistry Laboratories Tracy Lynn Haroff,2006

*The Inclusive Classroom* Margo A. Mastropieri,Thomas E. Scruggs,2007 The Inclusive Classroom Strategies for Effective Instruction Third Edition By Margo A Mastropieri and Thomas E Scruggs Recent Praise for The Inclusive Classroom Strategies for Effective Instruction Quite detailed Quite practical Everything is well linked to both research and standards Joseph E Nolan Indiana University of Pennsylvania This text is exceptional The authors have provided numerous practical examples throughout the book that are easy for inexperienced teachers to implement I believe even teachers with extensive experience would find new ideas in this textbook Karen B Clark Indiana University South Bend As you can see the real strength of The Inclusive Classroom is its emphasis on practical research based teaching and learning strategies in an inclusive environment The third edition focuses on the basic tools general educators need and directly relates content to the academic and professional demands of teachers in inclusive settings Practical and Effective Teaching and Learning Strategies In the Classroom features offer tips strategies and resources that address very specific need areas and can be practically applied to inclusive classroom situations Classroom Scenarios provide context for the specific teaching strategies

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*Laboratory Assessment in the Chemistry Classroom* Thomas Paul Hunt,1998 Holt World Geography Holt Rinehart & Winston,2007 Laboratory Manual [for] Fundamentals of Chemistry Fred H. Redmore,1979 *Prentice Hall Exploring Life Science* Anthea Maton,1997 **Chemistry, Student Study Guide** John A. Olmsted,Gregory M. Williams,2005-02-02

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