

UG (except BBA) CET Sample Paper

CHEMISTRY : (20 Questions)

1. If we take 44 gm. of CO_2 and 14 gm. of N_2 as a mixture, what is the mole fraction of CO_2 in the mixture?
 - A. 1/5
 - B. 1/3
 - C. 2/3
 - D. 1/4
2. According to Kinetic theory of gases, what will be the temperature when the rms velocity is 4 times of that at 300 K?
 - A. 300 K
 - B. 900 K
 - C. 4800 K
 - D. 1200 K
3. The enthalpy of vaporization of water is 186.5 J/mol. The entropy of vaporization is
 - A. 0.5 $\text{JK}^{-1}\text{mol}^{-1}$
 - B. 1.0 $\text{JK}^{-1}\text{mol}^{-1}$
 - C. 1.5 $\text{JK}^{-1}\text{mol}^{-1}$
 - D. 2.0 $\text{JK}^{-1}\text{mol}^{-1}$
4. If the solubility of Ag_2CrO_4 is S moles/litre , then its solubility product will be
 - A. S^2
 - B. S^3
 - C. $4S^3$
 - D. $2S^3$
5. The rate of the reaction becomes 4 times when the temperature is raised from 293 K to 313 K. The activation energy for such a reaction would be
 - A. 50.855 kJmol^{-1}
 - B. 52.849 kJmol^{-1}
 - C. 54.855 kJmol^{-1}
 - D. 56.855 kJmol^{-1}

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R Bogdan



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Table of Contents Imu Cet Sample Question Papers

1. Understanding the eBook Imu Cet Sample Question Papers
 - The Rise of Digital Reading Imu Cet Sample Question Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Imu Cet Sample Question Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Imu Cet Sample Question Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Imu Cet Sample Question Papers
 - Personalized Recommendations
 - Imu Cet Sample Question Papers User Reviews and Ratings
 - Imu Cet Sample Question Papers and Bestseller Lists
5. Accessing Imu Cet Sample Question Papers Free and Paid eBooks
 - Imu Cet Sample Question Papers Public Domain eBooks

- Imu Cet Sample Question Papers eBook Subscription Services
- Imu Cet Sample Question Papers Budget-Friendly Options
- 6. Navigating Imu Cet Sample Question Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Imu Cet Sample Question Papers Compatibility with Devices
 - Imu Cet Sample Question Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Imu Cet Sample Question Papers
 - Highlighting and Note-Taking Imu Cet Sample Question Papers
 - Interactive Elements Imu Cet Sample Question Papers
- 8. Staying Engaged with Imu Cet Sample Question Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Imu Cet Sample Question Papers
- 9. Balancing eBooks and Physical Books Imu Cet Sample Question Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Imu Cet Sample Question Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Imu Cet Sample Question Papers
 - Setting Reading Goals Imu Cet Sample Question Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Imu Cet Sample Question Papers
 - Fact-Checking eBook Content of Imu Cet Sample Question Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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