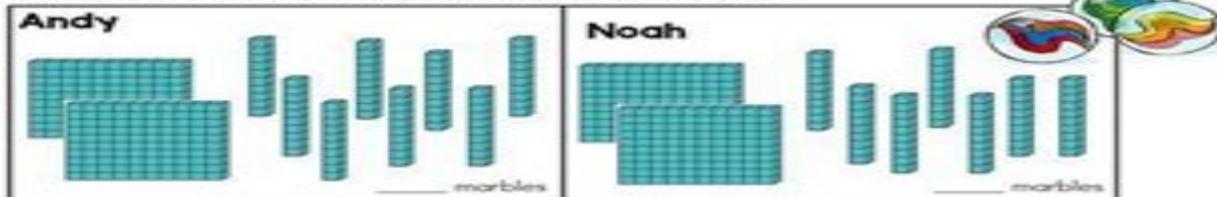


WHO IS RIGHT?

4

Think About It

Andy and Noah have marble collections. Noah told Andy, "Let me have 10 of your marbles and we will have the same amount." Andy thinks if he gives Noah 10 more marbles than Noah will have more than him. Who is right?



HIGHER ORDER THINKING

Place Value

THINK IT
Think About It
You and your class need to build a robot and they need to use 2 teams and they need to use the most parts. Which team will build the robot with the least amount of parts and which team will build the robot with the most parts?

Team A

Talk About It

WOULD YOU
Think About It
You want to take a fun trip for your family. You only want to travel 300 miles. Which place would you choose to visit? (Don't worry about the return trip.)

Location
Aquarium
Dinosaur Museum
Amusement Park
Zoo
Space Museum

Talk About It
Write About

WHO
Think About It
Andy and Noah have marbles. Noah told Andy, "Let me have 10 of your marbles and we will have the same amount." Andy thinks if he gives Noah 10 more marbles than Noah will have more than him. Who is right?

Andy

Talk About
Write About

WHICH IS RIGHT?
Think About It
Jack needs to represent 302 using base ten blocks. Which drawing below is correct?

A

B

Talk About It
Write About It



GRADE 2ND

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Lauren Gardner



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube

end c

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affected by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114.6-115.5 c thus revealing it to be acetanilide compound b had a melting point range of 122.6-124.5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

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