

1.2 Mole Conversions Solutions

PRACTICE EXAMPLES- To complete the practice examples (1) write the type of question you have, (2) write the equation with units, (3) show the cancellation of units and (4) calculate the solution.

1. Determine how many atoms are present in 2.5 moles of Silicon? (15.05×10^{23} atoms)

MOLES → ATOMS
$$\frac{2.5 \text{ moles}}{1} \times \frac{6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 15.05 \times 10^{23} \text{ atoms}$$

2. How many moles of Silicon is 3.01×10^{23} atoms? (.5 moles)

ATOMS → MOLES
$$\frac{3.011 \times 10^{23} \text{ atoms}}{1} \times \frac{1 \text{ mole}}{6.022 \times 10^{23} \text{ atoms}} = .5 \text{ moles}$$

3. What is the mass of 4 moles of Iron? (224 grams)

MOLES → GRAMS
$$\frac{4 \text{ moles}}{1} \times \frac{55.9 \text{ grams}}{\text{mole}} = 224 \text{ grams of C}$$

4. How many atoms are in 48.6 grams of Magnesium? (12.04×10^{23} atoms)

GRAMS → ATOMS
$$\frac{48.6 \text{ grams}}{1} \times \frac{1 \text{ mole}}{24.3 \text{ grams}} \times \frac{6.022 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 12.04 \times 10^{23} \text{ atoms}$$

5. How many grams are in 18.06×10^{23} atoms of Copper? (190.5 grams)

ATOMS → GRAMS
$$\frac{18.06 \times 10^{23} \text{ atoms}}{1} \times \frac{1 \text{ mole}}{6.022 \times 10^{23} \text{ atoms}} \times \frac{63.5 \text{ grams}}{1 \text{ mole}} = 190.5 \text{ grams}$$

6. How many moles of Nickel is 176 grams? (3 moles)

GRAMS → MOLES
$$\frac{176 \text{ grams of C}}{1} \times \frac{\text{mole}}{58.7 \text{ grams}} = 3 \text{ moles}$$

Mass Mole Conversion Handout Answer Key

Sally Solomon



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Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com gob

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elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

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SMITH,2012-02-16 eBook General Organic and Biological Chemistry 2e *Student's Guide, Chemistry, the Central Science*
James C. Hill,1991 *Basics for Chemistry* David A. Ucko,2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Basic Concepts of Chemistry Leo J. Malone,Theodore Dolter,2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance

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Chemistry Karen C. Timberlake,1992 Chemistry John McMurry,Robert C. Fay,2001 A colorful pedagogically enhanced standard textbook for the introductory course It begins with atomic structure proceeds next to bonding and molecules then to bulk physical properties of substances and ends with a study of chemical properties Each chapter concludes with a brief description of an interesting application or extension of the chapter subject a summary a list of key words and a large number of problems Many student oriented supplements are available Annotation copyright by Book News Inc Portland OR

Chemical Principles Peter Atkins,Loretta Jones,2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between

fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding

Chemistry James E. Brady, John R. Holm, 1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving

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General Chemistry Ralph H. Petrucci, 1989

Computerworld, 1999-01-18 For more than 40 years Computerworld has been the leading source of technology news and information for IT influencers worldwide Computerworld's award winning Web site Computerworld.com twice monthly publication focused conference series and custom research form the hub of the world's largest global IT media network

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