

## Electron Configuration Practice Worksheet

In the space below, write the unabbreviated electron configurations of the following elements:

- 1) sodium \_\_\_\_\_
- 2) iron \_\_\_\_\_
- 3) bromine \_\_\_\_\_
- 4) barium \_\_\_\_\_
- 5) neptunium \_\_\_\_\_

In the space below, write the abbreviated electron configurations of the following elements:

- 6) cobalt \_\_\_\_\_
- 7) silver \_\_\_\_\_
- 8) tellurium \_\_\_\_\_
- 9) radium \_\_\_\_\_
- 10) lawrencium \_\_\_\_\_

Determine what elements are denoted by the following electron configurations:

- 11)  $1s^2 2s^2 2p^6 3s^2 3p^4$  \_\_\_\_\_
- 12)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$  \_\_\_\_\_
- 13)  $[\text{Kr}] 5s^2 4d^{10} 5p^3$  \_\_\_\_\_
- 14)  $[\text{Xe}] 6s^2 4f^{14} 5d^5$  \_\_\_\_\_
- 15)  $[\text{Rn}] 7s^2 5f^{11}$  \_\_\_\_\_

Determine which of the following electron configurations are not valid:

- 16)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$  \_\_\_\_\_
- 17)  $1s^2 2s^2 2p^6 3s^3 3d^5$  \_\_\_\_\_
- 18)  $[\text{Ra}] 7s^2 5f^8$  \_\_\_\_\_
- 19)  $[\text{Kr}] 5s^2 4d^{10} 5p^5$  \_\_\_\_\_
- 20)  $[\text{Xe}]$  \_\_\_\_\_

# Electron Configuration Exam Questions

**Mina Lebitz**



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## **Table of Contents Electron Configuration Exam Questions**

1. Understanding the eBook Electron Configuration Exam Questions
  - The Rise of Digital Reading Electron Configuration Exam Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Electron Configuration Exam Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Electron Configuration Exam Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Electron Configuration Exam Questions
  - Personalized Recommendations
  - Electron Configuration Exam Questions User Reviews and Ratings
  - Electron Configuration Exam Questions and Bestseller Lists
5. Accessing Electron Configuration Exam Questions Free and Paid eBooks
  - Electron Configuration Exam Questions Public Domain eBooks
  - Electron Configuration Exam Questions eBook Subscription Services
  - Electron Configuration Exam Questions Budget-Friendly Options

6. Navigating Electron Configuration Exam Questions eBook Formats
  - ePub, PDF, MOBI, and More
  - Electron Configuration Exam Questions Compatibility with Devices
  - Electron Configuration Exam Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Electron Configuration Exam Questions
  - Highlighting and Note-Taking Electron Configuration Exam Questions
  - Interactive Elements Electron Configuration Exam Questions
8. Staying Engaged with Electron Configuration Exam Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Electron Configuration Exam Questions
9. Balancing eBooks and Physical Books Electron Configuration Exam Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Electron Configuration Exam Questions
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Electron Configuration Exam Questions
  - Setting Reading Goals Electron Configuration Exam Questions
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Electron Configuration Exam Questions
  - Fact-Checking eBook Content of Electron Configuration Exam Questions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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