

Practicing Electron Configuration

1. In the space below, write the **unabbreviated** electron configuration of the following elements.

- i. Sodium _____
- ii. Iron _____
- iii. Bromine _____
- iv. Barium _____
- v. Radium _____

2. In the space below, write the **abbreviated** electron configuration of the following elements.

- i. Cobalt _____
- ii. Silver _____
- iii. Tellurium _____
- iv. Radium _____
- v. Lawrencium _____

3. Determine what elements are denoted by the following electron configuration.

- i. $1s^2 2s^2 2p^6 3s^2 3p^4$: _____
- ii. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$: _____
- iii. $[\text{Kr}] 5s^2 4d^{10} 5p^3$: _____
- iv. $[\text{Xe}] 6s^2 4f^4 5d^6$: _____
- v. $[\text{Rn}] 7s^2 5f^{11}$: _____

4. Answer the following questions

- i. What is the shape of an s orbital? _____
- ii. How many s orbitals can there be in an energy level? _____
- iii. How many electrons can occupy an s orbital? _____
- iv. What is the shape of a p orbital? _____
- v. How many p orbitals can there be in an energy level? _____

Electron Configuration Review Answer Sheet

Fred M. Dewey



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Study Guide for Chemical Principles**

[by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

General Electric Review General Electric Company, 1921 *Physical Review*, 1929 Vols for 1903 include Proceedings of the American Physical Society *Study Guide, Chemical Principles, Fifth Edition, Zumdahl* Paul B. Kelter, Steven S.

Zumdahl, 2004-04 *Principles of Life* David M. Hillis, David E. Sadava, Richard Hill, Mary V. Price, 2014-07-15 With its first edition Principles of Life provided a textbook well aligned with the recommendations proposed in BIO 2010 Transforming Undergraduate Education for Future Research Biologists and Vision and Change in Undergraduate Biology Education Now Principles of Life returns in a thoroughly updated new edition that exemplifies the reform that is remaking the modern biology classroom

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molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives

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