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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect to zero energy at zero displacement from the spring s unstretched length $x = 0$ is $u = \frac{1}{2} kx^2$ figure 8 5 2 8 5 2 a glider between springs on an air track is an example of a horizontal mass spring system

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web figure 5 5 1 a potential energy diagram shows the total potential energy of a reacting system as the reaction proceeds a in an endothermic reaction the energy of the products is greater than the energy of the reactants and Δh is positive b in an

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web answer see below explanation the rate of a chemical reaction $k = A e^{-E_a/RT}$ is given in first approximation by $v = k[A]^m[B]^n$ where $k = A e^{-E_a/RT}$ where k depends by many factors as solvent geometry type of reaction E_a is the activation energy R is the universal constant of gases T is the absolute temperature

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web aug 8 2022 the energy changes that occur during a chemical reaction can be shown in a diagram called a potential energy diagram or sometimes called a reaction progress curve a potential energy diagram shows the change in potential energy of a system as reactants are converted into products the figure below shows basic potential energy

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