

# FIITJEE Talent Reward Exam - 2018

for students presently in

## Class 8

### Paper 1

Time: 3 Hours (9:30 am – 12:30 pm)

Code 8000

Maximum Marks: 210

#### Instructions:

Caution: Class, Paper, Code as given above **MUST** be correctly marked on the answer OMR sheet before attempting the paper. Wrong Class, Paper or Code will give wrong results.

- You are advised to devote 60 Minutes on Section-I, 60 Minutes on Section-II and 60 Minutes on Section-III.
- This Question paper consists of 3 sections. Marking scheme is given in table below:

Section	Subject	Question no.	Marking Scheme for each question	
			correct answer	wrong answer
SECTION – I	APTITUDE	1 to 30	+3	0
SECTION – II	PHYSICS (PART-A)	31 to 39	+2	0
	CHEMISTRY (PART-B)	40 to 48	+2	0
	MATHEMATICS (PART-C)	49 to 57	+2	0
	BIOLOGY (PART-D)	58 to 66	+2	0
SECTION – III	PHYSICS (PART-A)	67 to 78	+1	0
	CHEMISTRY (PART-B)	79 to 90	+1	0
	MATHEMATICS (PART-C)	91 to 102	+1	0
	BIOLOGY (PART-D)	103 to 114	+1	0

- Answers have to be marked on the OMR sheet. The Question Paper contains blank spaces for your rough work. No additional sheets will be provided for rough work.
- Blank papers, clip boards, log tables, slide rule, calculator, cellular phones, pagers and electronic devices, in any form, are not allowed.
- Before attempting paper write your OMR Answer Sheet No., Registration Number, Name and Test Centre in the space provided at the bottom of this sheet.

Note: Please check this Question Paper contains all 114 questions in serial order. If not so, exchange for the correct Question Paper.

OMR Answer Sheet No. : \_\_\_\_\_  
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# Fiitjee 8 Class Papers

**Roman Wölfel**



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the quantity of heat for a process is represented by the letter  $q$  the sign of  $q$  for an endothermic process is positive because the system is gaining heat a chemical reaction or physical

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one milligram is equal to 0 001 grams 790 drag each number to the correct location sort the numbers based on the number of significant figures they have one significant figure 1 000

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the quantities commonly encountered in chemical problems include the number of moles of a substance the number of atoms molecules or formula units of a substance and the mass in

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it is always a good idea to estimate the answer before you do the actual calculation in this case the mass given 35 00 g is less than the molar mass so the answer should be less than 1

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the atomic mass of an element expressed in grams is the mass of a mole of the element the mass of a mole of an element is the molar mass to calculate the molar mass of a compound

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the amount of a substance that contains  $6.02 \times 10^{23}$  representative particles of that substance the amount of a substance that contains  $6.02 \times 10^{23}$  representative particles of that

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web jul 27 2016 the plays on the call sheet came directly from our menu many of these plays would be repeated throughout this call sheet below is a screen shot of our call sheet with situational scripts the number of plays we had scripted for each situation was based on the number of times we would face that situation each game

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web mar 31 2021 during the covid impacted season of 2020 our first year as a staff at comstock park hs mi i noticed a disconnect between my thought process and how our offense should operate and the thought process of our jv offensive coordinator who was a first time play caller

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