

Name: _____

Date: _____

Student Expectation Circle

Learning Objectives: Students will be able to describe the process by which light is produced and emitted by atoms and molecules.

Learning Objectives: Students will be able to describe the process by which light is produced and emitted by atoms and molecules.

Prerequisite Knowledge: Students should understand the basic concepts of light.

Conceptual Model: Light is produced and emitted by atoms and molecules. In atoms, light is produced by the transition of an electron from a higher energy level to a lower energy level (excitation). In molecules, light is produced by the transition of a molecule from a higher energy state to a lower energy state.

1. Suppose a large light bulb is turned on. How do you think the light-emitting atoms and molecules behave?

The atoms and molecules are excited and emit light. The light is produced by the transition of an electron from a higher energy level to a lower energy level.

2. How do you think the light-emitting atoms and molecules behave when the light bulb is turned on?

The atoms and molecules are excited and emit light. The light is produced by the transition of an electron from a higher energy level to a lower energy level.

Learning Objectives:

Students will be able to describe the process by which light is produced and emitted by atoms and molecules. Students will be able to describe the process by which light is produced and emitted by atoms and molecules. Students will be able to describe the process by which light is produced and emitted by atoms and molecules.

1. How do you think the light-emitting atoms and molecules behave?

The atoms and molecules are excited and emit light.

2. How do you think the light-emitting atoms and molecules behave when the light bulb is turned on? How do you think the light-emitting atoms and molecules behave when the light bulb is turned on? How do you think the light-emitting atoms and molecules behave when the light bulb is turned on?

A. How do you think the light-emitting atoms and molecules behave when the light bulb is turned on?

The atoms and molecules are excited and emit light.

B. How do you think the light-emitting atoms and molecules behave when the light bulb is turned on?

The atoms and molecules are excited and emit light.



Gizmo Student Exploration Circuits Answer Key

RD Boyd



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