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Student Exploration: Electron Configuration

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."
2. Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."
3. In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? NO



Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? 3

2. A neutral atom has the same number of electrons and protons.

How many electrons are in a neutral lithium atom? 3

3. Select the ELECTRON CONFIGURATION tab, and check that **Energy** is selected. Click twice in the **1s** box at lower left and once in the **2s** box. Observe the atom model at right.

A. What do you see? a nucleus with somethings orbiting it



3	4	5	6	7	8
Sc	Ti	V	Cr	Mn	Fe
Y	Zr	Nb	Mo	Tc	Ru
La	Hf	Ta	W	Re	Os

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Table of Contents Gizmo Student Exploration Electron Configuration Answers

1. Understanding the eBook Gizmo Student Exploration Electron Configuration Answers
 - The Rise of Digital Reading Gizmo Student Exploration Electron Configuration Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploration Electron Configuration Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Gizmo Student Exploration Electron Configuration Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Student Exploration Electron Configuration Answers
 - Personalized Recommendations
 - Gizmo Student Exploration Electron Configuration Answers User Reviews and Ratings
 - Gizmo Student Exploration Electron Configuration Answers and Bestseller Lists
5. Accessing Gizmo Student Exploration Electron Configuration Answers Free and Paid eBooks
 - Gizmo Student Exploration Electron Configuration Answers Public Domain eBooks
 - Gizmo Student Exploration Electron Configuration Answers eBook Subscription Services

- Gizmo Student Exploration Electron Configuration Answers Budget-Friendly Options
- 6. Navigating Gizmo Student Exploration Electron Configuration Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Student Exploration Electron Configuration Answers Compatibility with Devices
 - Gizmo Student Exploration Electron Configuration Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploration Electron Configuration Answers
 - Highlighting and Note-Taking Gizmo Student Exploration Electron Configuration Answers
 - Interactive Elements Gizmo Student Exploration Electron Configuration Answers
- 8. Staying Engaged with Gizmo Student Exploration Electron Configuration Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploration Electron Configuration Answers
- 9. Balancing eBooks and Physical Books Gizmo Student Exploration Electron Configuration Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploration Electron Configuration Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Student Exploration Electron Configuration Answers
 - Setting Reading Goals Gizmo Student Exploration Electron Configuration Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Student Exploration Electron Configuration Answers
 - Fact-Checking eBook Content of Gizmo Student Exploration Electron Configuration Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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