

Honors Chemistry - Chapter 12 Test

True/False - Clearly indicate the correct choice

1. In an aqueous solution of table salt (NaCl), the table salt would be considered the solvent.
False
2. Water and gasoline are two liquids that would be considered miscible.
False
3. A negative sign listed for the enthalpy of solution means that energy is released in the formation of the solution.
True
4. A chemist finds that increasing the temperature leads to less of a substance dissolving in water. The substance is most likely a gas.
True
5. Salt or sugar dissolved in water is considered to be a homogeneous mixture.
True

Matching Set - Clearly indicate the correct choice

- a. solution b. colloid c. suspension
6. largest solute particle size c
 7. cannot be separated by filtration, will not scatter light a
 8. heterogeneous, does not separate on standing c
 9. particles settle out a
 10. smallest solute particle size b
 11. muddy water c
 12. shaving cream b

Multiple Choice - Clearly indicate the correct choice

13. A solute dissolved in alcohol is known as a/an
a. aqueous solution b. heterogeneous mixture c. tincture
d. immiscible solution
c
14. An electrolyte is a substance that dissolves in water to form a solution

Honors Chemistry Chapter Practice Test

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