

Flinn Scientific POGIL™ Activity

Isotopes

Are all atoms of an element alike?

Why?

The following activity will help you learn the important structural characteristics of an atom. How do we classify atoms? How does the combination of subatomic particles affect the mass and charge of an atom? What are isotopes? This is just a sampling of what we will address. Throughout this activity you will want to keep both Model 1 and a periodic table handy.

Model 1

Isotopes of Hydrogen			
Symbol	${}^1_1\text{H}$	${}^2_1\text{H}$	${}^3_1\text{H}$
Atomic Diagram with Name	<p style="text-align: center;">Hydrogen-1 (Protium)</p>	<p style="text-align: center;">Hydrogen-2 (Deuterium)</p>	<p style="text-align: center;">Hydrogen-3 (Tritium)</p>
Number of Protons	1	1	1
Number of Neutrons	0	1	2
Isotopes of Carbon			
Symbol	${}^{12}_6\text{C}$	${}^{13}_6\text{C}$	${}^{14}_6\text{C}$
Atomic Diagram with Name	<p style="text-align: center;">Carbon-12</p>	<p style="text-align: center;">Carbon-13</p>	<p style="text-align: center;">Carbon-14</p>
Number of Protons	6	6	6
Number of Neutrons	6	7	8
Isotopes of Magnesium			
Symbol	${}^{24}_{12}\text{Mg}$	${}^{25}_{12}\text{Mg}$	${}^{26}_{12}\text{Mg}$
Atomic Diagram with Name	<p style="text-align: center;">Magnesium-24</p>	<p style="text-align: center;">Magnesium-25</p>	<p style="text-align: center;">Magnesium-26</p>
Number of Protons	12	12	12
Number of Neutrons	12	13	14

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Isotopes Pogil Answer Key

Christian Drosten



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