

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



g.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

Br ₂	H ₂ O(g) + CO(g)	FeCl ₃ (aq) + I ₂ (aq)	HBr	H ₂
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4.5 Write down the TYPE of absorption in reaction II.

4.6 Compounds R and T are positional isomers.

The inorganic reagents shown below are available for reactions IV and V.

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IM Harris



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Table of Contents Grade Paper 2 Physical Science 2014 November December

1. Understanding the eBook Grade Paper 2 Physical Science 2014 November December
 - The Rise of Digital Reading Grade Paper 2 Physical Science 2014 November December
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Paper 2 Physical Science 2014 November December
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Paper 2 Physical Science 2014 November December
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Paper 2 Physical Science 2014 November December
 - Personalized Recommendations
 - Grade Paper 2 Physical Science 2014 November December User Reviews and Ratings
 - Grade Paper 2 Physical Science 2014 November December and Bestseller Lists
5. Accessing Grade Paper 2 Physical Science 2014 November December Free and Paid eBooks
 - Grade Paper 2 Physical Science 2014 November December Public Domain eBooks
 - Grade Paper 2 Physical Science 2014 November December eBook Subscription Services

- Grade Paper 2 Physical Science 2014 November December Budget-Friendly Options
- 6. Navigating Grade Paper 2 Physical Science 2014 November December eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade Paper 2 Physical Science 2014 November December Compatibility with Devices
 - Grade Paper 2 Physical Science 2014 November December Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Paper 2 Physical Science 2014 November December
 - Highlighting and Note-Taking Grade Paper 2 Physical Science 2014 November December
 - Interactive Elements Grade Paper 2 Physical Science 2014 November December
- 8. Staying Engaged with Grade Paper 2 Physical Science 2014 November December
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade Paper 2 Physical Science 2014 November December
- 9. Balancing eBooks and Physical Books Grade Paper 2 Physical Science 2014 November December
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Paper 2 Physical Science 2014 November December
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Paper 2 Physical Science 2014 November December
 - Setting Reading Goals Grade Paper 2 Physical Science 2014 November December
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Paper 2 Physical Science 2014 November December
 - Fact-Checking eBook Content of Grade Paper 2 Physical Science 2014 November December
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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