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### The Chemistry

Ionic, Covalent, and metallic compounds each have physical properties that relate to their structures, so there are some tests we can do to help identify the type of bonding in a substance.

### Melting Points

An ionic bond is made when ions of opposite charges come together to create a **crystal** lattice structure. In order to melt an ionic compound, the temperature has to rise until the ions move with enough force to break away from the oppositely charged ions around them.

Covalent Compounds don't break apart when they melt, so they only have to reach a temperature strong enough to pull them away from the other molecules to melt them. This is why most small covalent molecules are liquid or gas at room temperature.



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### Solubility & Conductivity



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### Solubility and Conductivity

Ionic compounds **dissolve** very easily in water to create an aqueous solution. Since water is a polar molecule, it attracts the ions and surrounds each one as the compound dissociates in the water. Once water is full of charged particles, it can conduct electricity very easily. Tap water conducts electricity because the chemicals used to treat the water are ions, so it makes the water into a conductor.



# Lesson 20 Covalent Bonds Unit 4

**L Reisser**



## **Lesson 20 Covalent Bonds Unit 4:**

## The Enigmatic Realm of **Lesson 20 Covalent Bonds Unit 4**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Lesson 20 Covalent Bonds Unit 4** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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