

6.1**ORGANIZING THE ELEMENTS****Section Review****Objectives**

- Explain how elements are organized in a periodic table
- Compare early and modern periodic tables
- Identify three broad classes of elements

Vocabulary

- periodic law
- metals
- nonmetals
- metalloids

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

- Chemists used the 1 of elements to sort them into groups. **1.** _____
- The periodic table organizes the elements into vertical 2 **2.** _____
- and horizontal 3 in order of increasing 4 . The table is **3.** _____
- constructed so that elements that have similar chemical properties **4.** _____
- are in the same 5 . 6 have a high luster, or sheen, **5.** _____
- when cut. Most nonmetals are 7 at room temperature. **6.** _____
- Elements with properties that are similar to those of metals **7.** _____
- and nonmetals are called 8 . Across the periodic table, **8.** _____
- the properties of elements become 9 metallic and **9.** _____
- 10 nonmetallic. **10.** _____

Lesson 6 1 Organizing The Elements Answer Key

JL Elias



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