

ISOTOPES & AVERAGE ATOMIC MASS WORKSHEET (3-PAGES)

includes answer key



NAME _____

Energy Worksheet

1. Complete the following table.

Element	Atomic Number	Mass Number	Number of Protons	Number of Neutrons	Number of Electrons
Hydrogen	1	1	1	0	1
Helium	2	4	2	2	2
Lithium	3	7	3	4	3
Boron	5	11	5	6	5

2. For each element determine the number of protons, neutrons, and electrons.

Element	Protons	Neutrons	Electrons
Hydrogen	1	0	1
Helium	2	2	2
Lithium	3	4	3

NAME _____

Isotopes Worksheet

1. Complete the following table.

Isotope	Atomic Number	Mass Number	Number of Protons	Number of Neutrons	Number of Electrons
^1_1H	1	1	1	0	1
^2_1H	1	2	1	1	1
^3_1H	1	3	1	2	1
$^{12}_6\text{C}$	6	12	6	6	6
$^{13}_6\text{C}$	6	13	6	7	6
$^{14}_6\text{C}$	6	14	6	8	6

2. For each isotope determine the number of protons, neutrons, and electrons.

Isotope	Protons	Neutrons	Electrons
^1_1H	1	0	1
^2_1H	1	1	1
^3_1H	1	2	1
$^{12}_6\text{C}$	6	6	6
$^{13}_6\text{C}$	6	7	6
$^{14}_6\text{C}$	6	8	6

NAME _____

Average Atomic Mass Worksheet

1. Calculate the average atomic mass for each element.

2. Calculate the average atomic mass for each element.

Element	Isotope	Abundance (%)	Mass (amu)
Hydrogen	^1_1H	99.98	1.0078
	^2_1H	0.02	2.0141
Carbon	$^{12}_6\text{C}$	98.93	12.0000
	$^{13}_6\text{C}$	1.07	13.0034

3. Calculate the average atomic mass for each element.

Isotopes And Average Atomic Mass Answer Key

Christopher John Willis



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