

General Chemistry – Unit 1 Worksheet 6

Dimensional Analysis

Use the factor-label method to make the following conversions. Remember to use the appropriate number of sf's in your answer.

Part 1

1. $74 \text{ cm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ meters
2. $8.32 \times 10^{-2} \text{ kg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ grams
3. $55.5 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ cm^3
4. $0.00527 \text{ cal} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ kilocalories
5. $9.52 \times 10^{-4} \text{ m} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ micrometers
6. $41.0 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ liters
7. $6.0 \times 10^{-1} \text{ g} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ mg
8. $8.34 \times 10^{-9} \text{ cg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ g
9. $5.0 \times 10^3 \text{ mm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ m
10. $1 \text{ day} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ seconds
11. $5 \times 10^4 \text{ mm} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ km
12. $9.1 \times 10^{-13} \text{ kg} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ ng
13. $1 \text{ year} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ hours (approximately)

General Chemistry Answer Key

Aftab S. Hassan



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General Chemistry Ralph H. Petrucci,F. Geoffrey Herring,Jeffrey D. Madura,Carey Bissonnette,2010-06-21 *General Chemistry* Darrell D. Ebbing,Larry K. Krannich,Joan I. Senyk,1996 **United States Naval Academy Admissions Regulations** ,1953 *Handbook of Academic Integrity* Sarah Elaine Eaton,2023-12-04 The book brings together diverse views from around the world and provides a comprehensive overview of academic integrity and how to create the ethical academy At the same time the Handbook does not shy away from some of the vigorous debates in the field such as the causes of academic integrity breaches There has been an explosion of interest in academic integrity in the last 20 30 years New technologies that have made it easier than ever for students to cut and paste coupled with global media scandals of high profile researchers behaving badly have resulted in the perception that plagiarism is on the rise This in combination with the massification and commercialisation of higher education has resulted in a burgeoning interest in the importance of academic integrity how to safeguard it and how to address breaches appropriately What may have seemed like a relatively easy topic to address students copying sources without attribution has in fact turned out to be a complex interdisciplinary field of research requiring contributions from linguists psychologists social scientists anthropologists teaching and learning specialists mathematicians accountants medical doctors lawyers and philosophers to name just a few Because of this broad interest and input this handbook serves as the single authoritative reference work which brings together the vast growing interdisciplinary and at times contradictory body of literature For both established researchers practitioners and those new to the field this Handbook provides a one stop shop as well as a launching pad for new explorations and discussions

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