

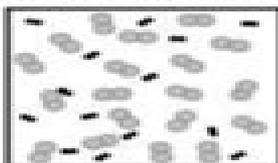
CHAPTER 18 HOW CHEMICALS MIX

Multiple Choice

Identify the choice that best completes the statement or answers the question.

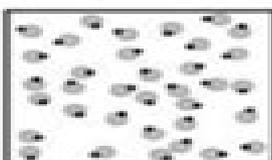
- _____ 1. A combination of two or more substances in which they no longer retain their chemical properties is called a(n) _____.
- heterogeneous mixture
 - compound
 - suspension
 - mixture
 - periodic trend

- _____ 2. The following image represents which kind of matter?



- a mixture
- an element
- a compound
- none of the above
- all of the above

- _____ 3. The following image represents which kind of matter?



- an element
- a compound
- a mixture
- none of the above
- all of the above

- _____ 4. Which of the following is a mixture?

- salt
- iron
- gold
- helium
- air

- _____ 5. How does a suspension differ from a solution?

- A suspension is a heterogeneous mixture consisting of different phases whereas a solution is a homogeneous mixture consisting of a single phase.
- The difference between a suspension and a solution can only be determined by chemical means.
- Although a solution and suspension are both homogeneous mixtures, only the components of a suspension will separate by spinning the mixture in a centrifuge.
- A suspension is a heterogeneous mixture whose components can be separated by simple filtration. A solution is a homogeneous mixture which cannot be separated by simple

How Chemicals Mix Guide Answers

Tao Wei



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