

## Calibrating a Standard Radioactive Isotope Decay Curve and Interpreting Data

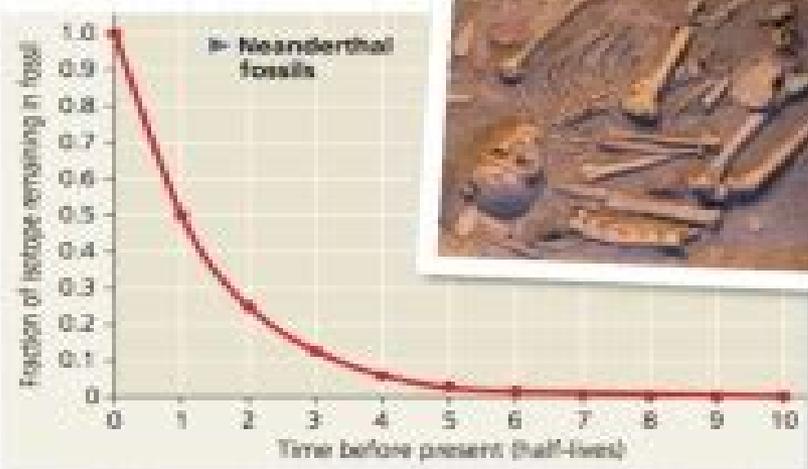
**When Did Neanderthals Become Extinct?** Neanderthals (*Homo neanderthalensis*) were living in Europe by 350,000 years ago, perhaps coexisting with early *Homo sapiens* in parts of Eurasia for hundreds or thousands of years. Researchers sought to more accurately determine the extent of their overlap by pinning down when Neanderthals became extinct. They used carbon-14 dating to determine the age of a Neanderthal fossil from the most recent uppermost archeological layer containing Neanderthal bones. In this exercise you will calibrate a standard carbon-14 decay curve and use it to determine the age of this Neanderthal fossil. The age will help you approximate the last time the two species may have coexisted at the site where this fossil was collected.

**How the Experiment Was Done** Carbon-14 ( $^{14}\text{C}$ ) is a radioactive isotope of carbon that decays to  $^{14}\text{N}$  at a constant rate.  $^{14}\text{C}$  is present in the atmosphere in small amounts at a constant ratio with both  $^{12}\text{C}$  and  $^{13}\text{C}$ , two other isotopes of carbon. When carbon is taken up from the atmosphere by a plant during photosynthesis,  $^{12}\text{C}$ ,  $^{13}\text{C}$ , and  $^{14}\text{C}$  isotopes are incorporated into the plant in the same proportions in which they were present in the atmosphere. These proportions remain the same in the tissues of an animal that eats the plant. While an organism is alive, the  $^{14}\text{C}$  in its body constantly decays to  $^{14}\text{N}$  but is constantly replaced by new carbon from the environment. **Once an organism dies, it stops taking in new  $^{14}\text{C}$  but the  $^{14}\text{C}$  in its tissues continues to decay,** while the  $^{12}\text{C}$  in its tissues remains the same because it is not radioactive and does not decay. Thus, scientists can calculate how long the pool of original  $^{14}\text{C}$  has been decaying in a fossil by measuring the ratio of  $^{14}\text{C}$  to  $^{12}\text{C}$  and comparing it to the ratio of  $^{14}\text{C}$  to  $^{12}\text{C}$  present originally in the atmosphere. The fraction of  $^{14}\text{C}$  in a fossil compared to the original fraction of  $^{14}\text{C}$  can be converted to years because we know that the half-life of  $^{14}\text{C}$  is 5,730 years—in other words, half of the  $^{14}\text{C}$  in a fossil decays every 5,730 years.

**Data from the Experiment** The researchers found that the Neanderthal fossil had approximately 0.0078 (or, in scientific notation,  $7.8 \times 10^{-3}$ ) as much  $^{14}\text{C}$  as the atmosphere. The questions will guide you through translating this fraction into the age of the fossil.

### Interpret the Data

1. A standard graph of radioactive isotope decay is shown at the top of the right column. **The graph line shows the fraction of the radioactive isotope over time (before present) in units of half-lives.** Recall that a half-life is the amount of time it takes for half of the radioactive isotope to decay. Labeling each data point with the corresponding fractions will help orient you to this graph. **Draw an arrow to the data point for half-life = 1 and write the fraction of  $^{14}\text{C}$  that will remain after one half-life.** Calculate the fraction of  $^{14}\text{C}$  remaining at each half-life and write the fractions on the graph near arrows pointing to the data points. Convert each fraction to a decimal number and round off to a maximum of three significant digits (zeros at the



beginning of the number do not count as significant digits). Also write each decimal number in scientific notation.

- Recall that  $^{14}\text{C}$  has a half-life of 5,730 years. To calibrate the x-axis for  $^{14}\text{C}$  decay, write the time before present in years below each half-life.
- The researchers found that the Neanderthal fossil had approximately 0.0078 as much  $^{14}\text{C}$  as found originally in the atmosphere. (a) Using the numbers on your graph, determine how many half-lives have passed since the Neanderthal died. (b) Using your  $^{14}\text{C}$  calibration on the x-axis, what is the approximate age of the Neanderthal fossil in years (round off to the nearest thousand)? (c) Approximately when did Neanderthals become extinct according to this study? (d) The researchers cite evidence that modern humans (*H. sapiens*) became established in the same region as the last Neanderthals approximately 39,000–42,000 years ago. What does this suggest about the overlap of Neanderthals and modern humans?
- Carbon-14 dating works for fossils up to about 75,000 years old; fossils older than that contain too little  $^{14}\text{C}$  to be detected. Most dinosaurs went extinct 65.5 million years ago. (a) Can  $^{14}\text{C}$  be used to date dinosaur bones? Explain. (b) Radioactive uranium-235 has a half-life of 704 million years. If it was incorporated into dinosaur bones, could it be used to date the dinosaur fossils? Explain.

 A version of this Scientific Skills Exercise can be assigned in MasteringBiology.

**Data from** S. Pinhasi et al., Revised age of late Neanderthal occupation and the end of the Middle Paleolithic in the northern Caucasus. *Proceedings of the National Academy of Sciences USA*, 147:8611–8616 (2015). doi:10.1073/pnas.1518938116

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