

A balloon full of air has a volume of 2.75 L at a temperature of 18°C. What is the balloon's volume at 45°C?

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

$$T_1 = 18 + 273 = 291 \text{ K}$$

$$T_2 = 45 + 273 = 318 \text{ K}$$

$$\frac{2.75 \text{ L}}{291 \text{ K}} = \frac{V_2}{318 \text{ K}}$$

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