

# Gas Laws Study Guide

Write the Law, Formula, Symbols, Units, and Examples for each of the following laws.

1. A sample of gas is heated and the volume increases.

2. The pressure of a gas is increased and the volume decreases.

3. A sample of gas is heated and the pressure increases.

4. A sample of gas is heated and the temperature increases.

5. The volume of a gas is increased and the pressure decreases.

6. A sample of gas is heated and the pressure decreases.

Name \_\_\_\_\_ Date \_\_\_\_\_

## Gas Laws Study Guide

Charles' Temperature  $T$  \_\_\_\_\_  $^{\circ}\text{C}$  \_\_\_\_\_  $\text{K}$

Boyle's Pressure  $P$  \_\_\_\_\_  $\text{atm}$  \_\_\_\_\_  $\text{torr}$   $P_1$  \_\_\_\_\_  $\text{atm}$   $P_2$  \_\_\_\_\_  $\text{atm}$

Gay-Lussac's \_\_\_\_\_  $T$  \_\_\_\_\_  $^{\circ}\text{C}$  \_\_\_\_\_  $\text{K}$

**Write the Formula for the Gas Laws**

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

Charles' Gas Law

$$P_{\text{Total}} = P_1 + P_2 + P_3$$

Boyle's Gas Law

$$P_1V_1 = P_2V_2$$

Boyle's Gas Law

$$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$$

Gay-Lussac's Gas Law

$$PV = nRT$$

Boyle's Gas Law

$$\frac{P_1}{T_1} = \frac{P_2}{T_2}$$

Gay-Lussac's Gas Law

Complete the following statements by filling in the blanks & circling the correct words.

As Pressure increases Volume \_\_\_\_\_ (increases/decreases) given that \_\_\_\_\_ is constant meaning that they have a direct/inverse relationship.

As Volume increases Temperature \_\_\_\_\_ (increases/decreases) given that \_\_\_\_\_ is constant meaning that they have a direct/inverse relationship.

As Pressure increases Temperature \_\_\_\_\_ (increases/decreases) given that \_\_\_\_\_ is constant meaning that they have a direct/inverse relationship.

As the number of moles of gas in a container increases the Volume \_\_\_\_\_ (increases/decreases), the Pressure \_\_\_\_\_ (increases/decreases), and the Temperature \_\_\_\_\_ (increases/decreases).

# Gas Laws Study Guide

**Lingsheng Yao**



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