

Unit 7 Gas Laws Test Review - Answer Key

1. A 0.4 L sample of gas is collected at a temperature of 45°C. At what temperature will the gas fill a 0.725 L container?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P_1 =$ $V_1 = 0.4 \text{ L}$ $T_1 = 318 \text{ K}$ $P_2 =$ $V_2 = 0.725 \text{ L}$ $T_2 = ?$	$\frac{V_1}{T_1} = \frac{V_2}{T_2}$	$\frac{0.4}{318} = \frac{.725}{T_2}$ $(.725 \times 318) / 0.4 = T_2$
answer with correct units: $T_2 = 576.38 \text{ K}$		

2. If 3 moles of a gas at a pressure of 3.5 atm has a volume of 110 L, what is the temperature?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P = 3.5 \text{ atm}$ $V = 110 \text{ L}$ $n = 3 \text{ mol}$ $R = 0.0821$ $T = ?$	$PV = nRT$	$(3.5 \times 110) = (3 \times 0.0821 \times T)$ $385 / .2463 = T$
answer with correct units: $T = 1563.13 \text{ K}$		

3. Identify if the following would cause an INCREASE or DECREASE in pressure in a gaseous system.

- Increase in the size of the container = **pressure decreases**
- More of the same gas is added to the container = **pressure increases**
- Increase in temperature = **pressure increases**
- A different gas is added to the container = **pressure increases**

4. A sealed container of Oxygen gas has a volume of 356 L. If heated from a temperature of 293-K to 318-K, what will the pressure be if the original pressure was 1.2 atm?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P_1 = 1.2 \text{ atm}$ $V_1 =$ $T_1 = 293 \text{ K}$ $P_2 = ?$ $V_2 =$ $T_2 = 318 \text{ K}$	$\frac{P_1}{T_1} = \frac{P_2}{T_2}$	$\frac{1.2}{293} = \frac{P_2}{318}$ $(1.2 \times 318) / 293 = P_2$

Gas Laws Test Review Answers

Zhe Xu



Gas Laws Test Review Answers:

Whispering the Techniques of Language: An Mental Journey through **Gas Laws Test Review Answers**

In a digitally-driven earth where monitors reign supreme and instant interaction drowns out the subtleties of language, the profound techniques and emotional nuances concealed within words often go unheard. Yet, nestled within the pages of **Gas Laws Test Review Answers** a charming fictional treasure pulsing with fresh feelings, lies a fantastic journey waiting to be undertaken. Published by a skilled wordsmith, this enchanting opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within the material of every word. Within the psychological depths of this touching review, we can embark upon a honest exploration of the book is core themes, dissect their interesting writing model, and fail to the strong resonance it evokes deep within the recesses of readers hearts.

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