

FNP:(Family Nurse Practitioner) Exam Study Guide

Trauma to Kiesselbachs plexus:

(Ans- **Will result in an anterior nosebleed**)

The diagnostic or gold standard test for sickle cell anemia, G6PD anemia, and alpha or beta thalassemia:

(Ans- **Hemoglobin electrophoresis**)

Erythromycin "allergy" vs adverse reaction:

(Ans- **Adverse reaction-symptoms of nausea or GI upset
allergy -hives, angioedema**)

Acute mononucleosis:

(Ans- **Pt will most likely be a teen presenting w classic triad of sore throat,
prolonged fatigue, and enlarged cervical nodes.**)

Alpha thalassemia:

(Ans- **More common among Southeast Asians such as, Indian, Chinese, or
Filipino descent**)

Lupus

(Ans- **Malar rash(butterfly rash)Pts need to avoid or to minimize sunlight
exposure(photosensitivity)**)

Tx polymyalgic rheumatica(PMR)

(Ans- **1st line tx for PMR includes long-term steroids. Long term steroids
are commonly used to control symptoms(pain, stiffness on shoulders, and
hip girdle). PMR pts have higher risk for temporal arteritis.**)

Gold standard exam for temporal arteritis:

(Ans- **Biopsy of the temporal artery, refer pt to ophthalmologist for mgt**)

Fnp Study Guide

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Fnp Study Guide:

How to Study for Your Fnp Certification the Quick Way. Tina Shropshire, 2018-05-14 Tired of reading thick text books that have more information in them than you know what to do with or can even process Are you ready to read a book that will get straight to the point of what will actually be on your exam If so this book is for you This is a quick study guide specifically designed to help you study for the FNP boards without the over kill Family Nurse Practitioner Certification Intensive Review, Third Edition Maria T. Codina Leik, 2017-11-28 Note to Readers Publisher does not guarantee quality or access to any included digital components if book is purchased through a third party seller Includes 6 months free ExamPrepConnect Digital Access with print purchase Doody's Medical Review 100 Points 5 Stars First Edition The third edition of this highly acclaimed review continues to promote efficient time saving study by synthesizing the complex knowledge needed to pass the exam into a concise well organized format It is updated with key information about the new ANCC and AANPCB certification exams including new test format questions featuring photos of skin and eye conditions and EKGs and more challenging drag and drop and multiple choice questions Also new to the third edition is more detailed information about nonclinical topics on the exam including research ethics medico legal issues advanced practice law health practices of various cultures and reimbursement guidelines One hundred new review questions throughout the book and on the practice exam including answer key and rationale increase the total number of questions to more than 700 Designed to help FNP students boost their confidence through intensive review and high quality questions the third edition features succinct precisely targeted need to know details of diseases commonly seen in primary care It covers the entire life span from pediatrics through geriatrics and offers extensive test taking techniques and question dissection and analysis sections that provide strategies for identifying the best clues during the problem solving process Organized by body system sections are consistently formatted and encompass Danger Signals Normal Findings Lab Findings Benign Variants and individual Disease Review topics for each system Additional outstanding features include Exam Tips and Clinical Pearls that help to focus system study on key information NEW TO THE THIRD EDITION Updated information reflecting the new FNP certification exam New test format questions used by the ANCC including photos of skin and eye conditions EKGs and more complex drag and drop and multiple choice questions 100 new in depth review questions for a total of more than 700 Differential Diagnosis tables Additional information about nonclinical topics including culture spirituality and health practices evidence based medicine and more KEY FEATURES Provides a succinct precisely targeted review of diseases commonly seen in primary care Delineates strategic question dissection techniques for study success Includes Exam Tips and Clinical Pearls to help prioritize and interpret test content Outlines Normal Findings and Benign Variants in physical assessment of each body system Offers intensive pharmacology review Content available digitally via ExamPrepConnect for study on the go ExamPrepConnect Features All the high quality content from the book Personalized study plan based on

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solution the higher the K_{sp} the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized $Ca(OH)_2$ when calcium is put in water reacts as follows $Ca(OH)_2 \rightleftharpoons Ca^{2+} + 2OH^-$

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web K_{sp} of calcium hydroxide 1 author j m mccormick last update october 13 2013 introduction an equilibrium constant K is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of K

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web sep 18 2020 in this experiment we will titrate a saturated calcium hydroxide solution with a standardized hcl solution we will determine the hydroxide ion concentration for the saturated calcium hydroxide

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the K_{sp} of the compound you will do this by titrating the prepared $Ca(OH)_2$ solution with a standard hydrochloric acid solution

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web in today's experiment you will determine the solubility product K_{sp} of calcium hydroxide $Ca(OH)_2$ by measuring the concentration of $Ca(OH)_2$ in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the K_{sp} of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of $Ca(OH)_2$ is represented in equation form as shown below $Ca(OH)_2(s) \rightleftharpoons Ca^{2+}(aq) + 2OH^-(aq)$ the solubility product expression describes in mathematical terms the equilibrium

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web K_{sp} for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a pH regulating agent notice that calcium hydroxide is divalent and thus has twice

K_{sp} calcium hydroxide chem132 determination of the K_{sp} of calcium - Jun 24 2022

web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $Ca(OH)_2 \rightleftharpoons Ca^{2+} + 2OH^-$ $5 \times 10^{-2} M$ calcium ions $5 \times 10^{-2} M$ $2 \times 10^{-2} M$ hydroxide ions calculate calcium hydroxide K_{sp} using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the K_{sp} of a slightly soluble basic salt and how solubility changes with temperature K_{sp} is another form of equilibrium expression that deals with solubility the reaction that will take place is $Ca(OH)_2 \rightleftharpoons Ca^{2+} + 2OH^-$

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web question experiment 8 determining K_{sp} for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 HCl molarity 0.050 M volume of filtered $Ca(OH)_2$ solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml OH^- in saturated $Ca(OH)_2$ M Ca^{2+} in saturated $Ca(OH)_2$ M K_{sp} of $Ca(OH)_2$ average K_{sp} of $Ca(OH)_2$

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web the chemical equation used was $Ca(OH)_2(s) \rightleftharpoons Ca^{2+}(aq) + 2OH^-(aq)$ the solubility product was determined to be $K_{sp} Ca^{2+} OH^-$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant K_{sp} chemistry libretxts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{sp} Ca^{2+}_3 PO_4^{3-}_2 = [Ca^{2+}]^3 [PO_4^{3-}]^2$

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web lab report 8 determining K_{sp} of calcium hydroxide the purpose of this was to titrate a saturated $Ca(OH)_2$ solution with a standard HCl solution and find the K_{sp} or the OH^- concentration was found and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochloric acid

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web question determining the K_{sp} of calcium hydroxide $Ca(OH)_2 \rightleftharpoons Ca^{2+}(aq) + 2OH^-(aq)$ for this experiment the titration of calcium hydroxide with hydrochloric acid was investigated determining the K_{sp} of calcium hydroxide $Ca(OH)_2 \rightleftharpoons Ca^{2+}(aq) + 2OH^-(aq)$ for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the K_{sp} of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of $Ca(OH)_2$ is represented in equation form lution of $Ca(OH)_2$ on ionic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest K_{sp} value to the theoretical K_{sp} for $Ca(OH)_2$ was the determination through pH method 2 introduction calcium hydroxide

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web calculate the ksp for calcium hydroxide $\text{Ca}^{2+} \text{OH}^-$ 2.0×10^{-2} ksp 5×10^{-4} find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}

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