



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Two Elements
 (Name the more electronegative element first)

- EXAMPLE**
- CO2
 Carbon Dioxide (Naming Periodic)
- Identifying the naming**
1. Name the more electronegative element first.
 a. Oxygen is more electronegative than carbon so oxygen is named first.
 b. The name of oxygen is oxygen.
 2. Name the less electronegative element as the second element.
 a. Carbon is less electronegative than oxygen so carbon is named second.
 b. The name of carbon is carbon.
 3. The name of the compound is oxygen carbon dioxide.

Binary Compounds	
First Element	Prefix
1	mono-
2	di-
3	tri-
4	tetra-
5	penta-
6	hexa-
7	hepta-
8	octa-
9	nona-
10	deca-

Common Polyatomic Ions	
Formula	Name
<chem>OH-</chem>	hydroxide
<chem>NO3-</chem>	nitrate
<chem>SO4^2-</chem>	sulfate
<chem>CO3^2-</chem>	carbonate
<chem>PO4^3-</chem>	phosphate
<chem>NO2-</chem>	nitrite
<chem>SO3^2-</chem>	sulfite
<chem>PO3^3-</chem>	phosphite
<chem>NO-</chem>	nitrosyl
<chem>NO2+</chem>	nitrosonium

Naming Compounds with a Metal and a Nonmetal

- EXAMPLE**
- CaCl2
 Calcium Chloride
- Identifying the naming**
1. Name the metal first (the cation).
 a. Calcium is the metal so calcium is named first.
 b. The name of calcium is calcium.
 2. Name the nonmetal second (the anion).
 a. Chlorine is the nonmetal so chlorine is named second.
 b. The name of chlorine is chloride.

Naming Anions			
Element	Element Name	Element	Element Name
Fluorine	fluorine	Fluoride	fluoride
Oxygen	oxygen	Oxide	oxide
Sulfur	sulfur	Sulfide	sulfide
Chlorine	chlorine	Chloride	chloride
Bromine	bromine	Bromide	bromide
Iodine	iodine	Iodide	iodide
Carbon	carbon	Carbide	carbide
Nitrogen	nitrogen	Nitride	nitride

Common Names for Specific Compounds			
Formula	Common Name	Formula	Common Name
<chem>NaCl</chem>	table salt	<chem>NaOH</chem>	caustic soda
<chem>CaCl2</chem>	plaster	<chem>CaCO3</chem>	limestone
<chem>NaNO3</chem>	saltpetre	<chem>CaSO4</chem>	gypsum
<chem>Na2CO3</chem>	soda ash	<chem>Na2SO4</chem>	glauber salt

Naming Compounds Study Guide

Michal Rosen-Zvi



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