

Nomenclature Skill Review: Naming Binary Compounds

Binary Ionic Compounds (Type I & II) & Review rules & for look at flowchart for guidance

Write the symbol for the ion:	Name the ion:
Lithium	Li^+
Strontium	Sr^{2+}
Bromide	Br^-
Cobalt (III)	Co^{3+}
Oxide	O^{2-}
Calcium	Ca^{2+}
Copper (II)	Cu^{2+}
Aluminum	Al^{3+}
Silver	Ag^+
Vanadium (V)	VO_4^{3-}
Write the formula:	Name the compound:
Cesium chloride	LiBr
Calcium fluoride	$\text{Hg}(\text{NO}_2)_2$
Rubidium sulfide	K_2S
Copper (II) oxide	SnO
Potassium iodide	Fe_2O_3
Titanium (IV) selenide	Ni_2CO_3
Silver sulfide	BaSe
aluminum sulfate	HgF
Gallium phosphate	$\text{Pb}(\text{ClO}_4)_2$
Beryllium nitride	ZnTe

Binary Molecular Compounds (Type III) & Review rules & for look at flowchart for guidance

Write the formula:	Write the formula:	Name the compound:
2	Diboron hexafluoride	P_2O_5
2	Carbon monoxide	As_2S_5
3	Phosphorus pentabromide	N_2O
4	Trinitrogen heptafluoride	SeCl_4
5	Tetraarsenic decoxide	Si_4Br_7
7	Sulfur dioxide	S_2F_6
8		
9		
10		

Naming Review Guide Answer Key

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