

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____ Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

| Atomic Radius (pm) | | | | | | | |
|--------------------|-----|-----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 152 | Sodium - 11 | 190 | Potassium - 19 | 243 | Rubidium - 27 | 295 |
| Beryllium - 4 | 112 | Magnesium - 12 | 145 | Calcium - 20 | 194 | Strontium - 28 | 219 |
| Boron - 5 | 87 | Aluminum - 13 | 118 | Gallium - 31 | 135 | Indium - 40 | 156 |
| Carbon - 6 | 67 | Silicon - 14 | 111 | Germanium - 32 | 125 | Tin - 50 | 145 |
| Nitrogen - 7 | 55 | Phosphorus - 15 | 98 | Arsenic - 33 | 114 | Antimony - 51 | 133 |
| Oxygen - 8 | 48 | Sulfur - 16 | 88 | Selenium - 34 | 103 | Tellurium - 52 | 123 |
| Fluorine - 9 | 42 | Chlorine - 17 | 79 | Bromine - 35 | 94 | Iodine - 53 | 115 |

| Ionization Energy (kJ/mol) | | | | | | | |
|----------------------------|-----|----------------|-----|----------------|-----|----------------|-----|
| Lithium - 3 | 520 | Sodium - 11 | 496 | Potassium - 19 | 419 | Rubidium - 27 | 403 |
| Beryllium - 4 | 900 | Magnesium - 12 | 738 | Calcium - 20 | 590 | Strontium - 28 | 550 |

Atomic Radius



Graphing Periodic Trends Lab Answer Key

Ying-Ying Zheng



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Laboratory Manual Jo A. Beran,1990 *Computer Based Projects for a Chemistry Curriculum* Thomas J.

Manning,Aurora P. Gramatges,2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom

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Table of Contents Graphing Periodic Trends Lab Answer Key

1. Understanding the eBook Graphing Periodic Trends Lab Answer Key
 - The Rise of Digital Reading Graphing Periodic Trends Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Periodic Trends Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Graphing Periodic Trends Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Periodic Trends Lab Answer Key
 - Personalized Recommendations
 - Graphing Periodic Trends Lab Answer Key User Reviews and Ratings
 - Graphing Periodic Trends Lab Answer Key and Bestseller Lists
5. Accessing Graphing Periodic Trends Lab Answer Key Free and Paid eBooks
 - Graphing Periodic Trends Lab Answer Key Public Domain eBooks
 - Graphing Periodic Trends Lab Answer Key eBook Subscription Services
 - Graphing Periodic Trends Lab Answer Key Budget-Friendly Options
6. Navigating Graphing Periodic Trends Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Graphing Periodic Trends Lab Answer Key Compatibility with Devices
 - Graphing Periodic Trends Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Graphing Periodic Trends Lab Answer Key
 - Highlighting and Note-Taking Graphing Periodic Trends Lab Answer Key
 - Interactive Elements Graphing Periodic Trends Lab Answer Key
8. Staying Engaged with Graphing Periodic Trends Lab Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Graphing Periodic Trends Lab Answer Key
9. Balancing eBooks and Physical Books Graphing Periodic Trends Lab Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Graphing Periodic Trends Lab Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Graphing Periodic Trends Lab Answer Key
- Setting Reading Goals Graphing Periodic Trends Lab Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Graphing Periodic Trends Lab Answer Key
- Fact-Checking eBook Content of Graphing Periodic Trends Lab Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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