

GRAPHING PERIODIC TRENDS

Standard: Students know how to use the periodic table to identify trends in ionization energy, electronegativity, and the relative sizes of ions and atoms.

PRE-LAB DISCUSSION:

The Periodic Table is arranged according to the Periodic Law. The Periodic Law states that when elements are arranged in order of increasing atomic number, their physical and chemical properties show a periodic pattern. Students can discover these patterns by examining the changes in properties of elements on the Periodic Table. The properties that will be examined in this lesson are: atomic radius AND first ionization energy.

PURPOSE:

To understand periodic trends relating to atomic radius, ionization energy, and electronegativity.

PROCEDURE:

Use the information in these tables to complete the graph as described below.

RESULTS:

Symbol	Atomic Radius (Picometers)	First Ionization (Kilojoules/mole)	Electronegativity (A-point scale)
H	37	1312	2.1
Li	128	520	1.0
Na	190	495	0.9
K	203	419	0.8
Rb	220	403	0.8
Cs	244	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization (Kilojoules/mole)	Electronegativity (A-point scale)
Na	190	495	0.9
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	104	1000	2.0
Cl	102	1251	3.0
Ar	100	1521	—

1. Define each of these terms: Atomic radius, First Ionization Energy, and Electronegativity
2. Using colored pencil or pen, list under the symbol (in this order) the Atomic Radius, First Ionization Energy, and Electronegativity. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First Ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)
5. There is NO ABSTRACT to write for this activity :)

Graphing Periodic Relationships Lab Answers

SJ Ball



Graphing Periodic Relationships Lab Answers:

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Table of Contents Graphing Periodic Relationships Lab Answers

1. Understanding the eBook Graphing Periodic Relationships Lab Answers
 - The Rise of Digital Reading Graphing Periodic Relationships Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Periodic Relationships Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Graphing Periodic Relationships Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Periodic Relationships Lab Answers
 - Personalized Recommendations
 - Graphing Periodic Relationships Lab Answers User Reviews and Ratings
 - Graphing Periodic Relationships Lab Answers and Bestseller Lists
5. Accessing Graphing Periodic Relationships Lab Answers Free and Paid eBooks
 - Graphing Periodic Relationships Lab Answers Public Domain eBooks
 - Graphing Periodic Relationships Lab Answers eBook Subscription Services
 - Graphing Periodic Relationships Lab Answers Budget-Friendly Options

6. Navigating Graphing Periodic Relationships Lab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Graphing Periodic Relationships Lab Answers Compatibility with Devices
 - Graphing Periodic Relationships Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Graphing Periodic Relationships Lab Answers
 - Highlighting and Note-Taking Graphing Periodic Relationships Lab Answers
 - Interactive Elements Graphing Periodic Relationships Lab Answers
8. Staying Engaged with Graphing Periodic Relationships Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Graphing Periodic Relationships Lab Answers
9. Balancing eBooks and Physical Books Graphing Periodic Relationships Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Graphing Periodic Relationships Lab Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Graphing Periodic Relationships Lab Answers
 - Setting Reading Goals Graphing Periodic Relationships Lab Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Graphing Periodic Relationships Lab Answers
 - Fact-Checking eBook Content of Graphing Periodic Relationships Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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