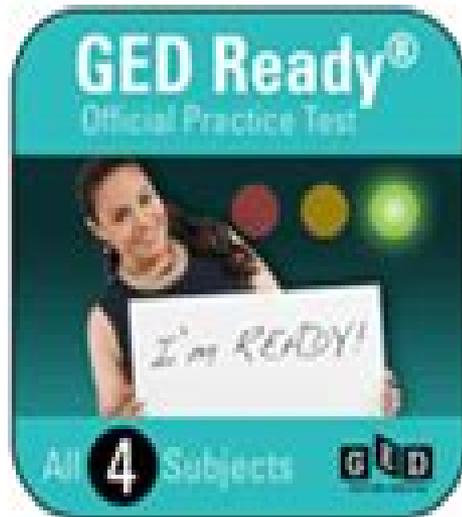


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web dibenzalacetone was synthesized in the lab from benzaldehyde and acetone under basic conditions the gc ms was used to relate the molar mass of the product thin layer chromatography was used to visualize the relative rf factor for each isomer uv vis instrument showed three peaks that were assigned to each isomer

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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web oct 25 2013 experiment performed on may 6 and 8 2003 br the purpose strong of strong this experiment was to synthesize dibenzalacetone via an aldol condensation reaction br between acetone and benzaldehyde this was done by mixing the two reactants with naoh and ethanol br

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a double cross aldol claisen schmidt condensation between excess benzaldehydehydroxide ethanol solution was utilized to
produce dibenzalacetone

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web dibenzalacetone a introduction in an aldol condensation two molecules of aldehyde or ketone are joined together along
with the loss of water for example in the aldol condensation shown in figure 1 two molecules of ethanal are reacted to
produce an a b unsaturated aldehyde and water figure 1 aldol condensation of ethanal to