

Microbiology Exam 2

Chapters 7, 13, 8, 20, 15

CHAPTER 7: THE CONTROL OF MICROBIAL GROWTH

- Chemical Methods of Microbial Control
 - Disinfection and sterilization
 - Enveloped viruses are more susceptible to sterilization
 - Less reliable than heat, but is more accessible
 - Sterilants: used to sterilize instruments
 - Disinfectants
 - High level disinfectants: reduce number of microorganisms to great extent (instruments)
 - Intermediate level disinfectant: used on stethoscope and things used for outpatient testing that don't have time to be fully sterilized
 - Low level disinfectant: used every day in kitchen/bathrooms
 - How do we select a germicidal chemical?
 - Is it toxic to humans?
 - Can it withstand the presence of organic matter?
 - Is it compatible with different materials?
 - Does it leave a residue? If yes has to be rinsed with sterile water
 - Cost and availability (effective=expensive)
 - Storage and stability (most have a strict shelf life and must be diluted to be stable)
 - Environmental risk (disposal impact on water/environment)
 - Phenols (carbolic acid) and Phenolics: very effective in reducing infection risk, but very corrosive
 - Phenolic: derivative of a phenol (chemically altered to reduce irritating qualities)
 - Lysol is a phenolic (family is cresols, derived from coal tar)
 - Phenols: controls odor in sewage, irritates skin and has bad odor
 - Bisphenols
 - Derivatives of phenol that contain two phenolic groups
 - Hexachlorophane: used for degerming, prevents staph used in nurseries with caution because it causes neurological problems
 - Biguanides
 - Usually effective against gram + bacteria, and some gram - (not pseudomonas)
 - Chlorhexidine: effective on skin and mucus membrane (includes mouthwash and topical creams)
 - Halogens (antimicrobial agents)
 - Chlorine (Water treatment and gas): hypochlorous acid (disinfectant), calcium hypochlorite (cleans utensils in food industry), sodium hypochlorite (bleach), Chlorine Dioxide (used to clean water), Chloramines (chlorine and ammonium to treat water and clean)
 - Iodine (antiseptic): Betadine (combined with organic molecule to allow slow release over time), Tincture (combined with alcohol that causes pain and stains skin brown)
 - Alcohols: disinfectant and antiseptic
 - Effectively kill bacteria and fungi, not endospores/non-enveloped viruses
 - Coagulation of proteins, enzymes, and damage membranes
 - Used to enhance other chemicals (works better when combined) Isopropanol (rubbing alcohol)
 - Heavy Metal and their compounds
 - Silver (very effective for burn victims, pseudomonas), Mercury, copper, zinc (most are too toxic to use and inhibit growth of bacteria)
 - Surface Active agents/Surfactants reduce surface tension in molecules of a liquid
 - soaps, detergents
 - Acid-Anionic Sanitizers: used to clean utensils in food industry(have a negative charge that doesn't work w/ microbes)
 - Quaternary Ammonium Compounds (QUATS): used in mouthwash as a mild antiseptic
 - Cepacol, Zephiran

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