

Name _____

KEY

Period _____

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REVIEW QUESTIONS FOR FINAL EXAM (2nd Semester)

SECTION 12.1

Heat Transfer: $Q = mC\Delta T$ or $Q = mC(T_f - T_i)$

Specific Heat of Common Substances			
Material	Specific Heat $J/kg \cdot ^\circ C$	Material	Specific Heat $J/kg \cdot ^\circ C$
aluminum	900	lead	130
brass	376	nickel	210
carbon	712	silver	235
copper	385	iron	450
glass	840	mercury	140
ice	2090	oil	200
iron	450		

1) Two solid objects are in contact as shown below.



Heat will always flow from object A to object B if object B has a lower

- a) mass
- b) total energy
- c) specific heat
- d) temperature
- e) none of the answers

For questions 2-3: The diagram below shows two solids with their temperatures (A=50°C, B=80°C).



2) What occurs when the two solids are placed in contact with each other as shown?

- a) Heat energy flows from A to B. Solid A cools down.
- b) Heat energy flows from A to B. Solid A gets hotter.
- c) Heat energy flows from B to A. Solid B cools down.
- d) Heat energy flows from B to A. Solid B gets hotter.

3) The net flow of heat energy will stop from one object to the other when they both have the same

- a) energy
- b) temperature
- c) specific heat
- d) all the answers are correct

4) How much heat is transferred when a 0.2 kg piece of copper's temperature is raised from 50°C to 80°C?

$$Q = mC\Delta T$$

$$Q = (0.2 \text{ kg})(385 \text{ J/kg} \cdot ^\circ \text{C})(80^\circ \text{C} - 50^\circ \text{C}) = 2310 \text{ J}$$

5) A 0.5 kg block of lead cools down from 30 °C to 20 °C. How much heat was transferred?

$$Q = (0.5 \text{ kg})(130 \text{ J/kg} \cdot ^\circ \text{C})(20^\circ \text{C} - 30^\circ \text{C})$$

$$Q = -650 \text{ J}$$

6) The heat transfer of a substance is 9400 J. The mass of this substance is 2 kg and its temperature rises from 80 °C to 100 °C. Based on the specific heat that you will calculate for this substance, what is it?

$$C = \frac{Q}{m\Delta T} = \frac{9400 \text{ J}}{(2 \text{ kg})(100^\circ \text{C} - 80^\circ \text{C})} = 235 \text{ J/kg} \cdot ^\circ \text{C}$$

SILVER

7) From the choices below, which material would be better for a pan handle?

- a) brass 376
- b) silver 235
- c) copper 385
- d) iron 450
- e) zinc 388
- f) lead 130

8) Provide a brief description of absolute zero and their temperatures in Kelvin (K) and degrees Celsius (°C).

The lowest temperature when all particle motion stops. $T_C = -273^\circ \text{C}$, $T_K = 0 \text{ K}$.

9) The boiling point of water is 100 °C. The boiling point of water in Kelvin (K) will be

$$T_K = 273 + T_C = 273 + 100 = 373 \text{ K}$$

10) A 1-liter of water taken out of a 5-gallon bucket full of water at 20 °C will

- a) register a lower temperature than that of the water left in the bucket
- b) have less thermal energy than that of the water left in the bucket
- c) have less specific heat than that of the water left in the bucket

11) 243 K in degrees Celsius (°C) is

$$T_C = T_K - 273 = 243 \text{ K} - 273 = -30^\circ \text{C}$$

12) The kinetic molecular theory can explain the existence of thermal energy in a substance.

- a) true
- b) false

13) When using a thermometer for measuring the temperature of a substance, one should record the temperature

- a) about 2 minutes after the thermometer has been placed in the substance. That is usually the time needed for the liquid inside to expand or shrink completely.
- b) until thermal equilibrium has been reached
- c) until all the heat from the substance has been transferred to the thermometer
- d) until the thermal energy from the substance is equal to the thermal energy of the thermometer
- e) all of the answers

14) A calorimeter contains 1 kg of water at 20 °C. A 0.20 kg block of zinc at 120 °C is placed in the water. What is the approximate final temperature of the system?

$$T_f = \frac{m_A C_A T_{Ai} + m_B C_B T_{Bi}}{m_A C_A + m_B C_B}$$

$$T_f = \frac{(1)(4180)(20^\circ \text{C}) + (0.2)(388)(120^\circ \text{C})}{(1)(4180) + (0.2)(388)}$$

$$T_f = \frac{83600 + 9312}{4257.6} = 21.8^\circ \text{C}$$

Final Exam Review Physics Answer Key

Clifford Lane



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sometimes be a bit tricky to write this letter have a look at the sample letters given below to get an idea of how to write the same sample letter 1 change of address letter to the bank following marriage sana shaikh humayun society

request letter for change of home branch sample letter to bank - Oct 06 2023

web jun 29 2021 courteously i am your name having type of bank account account in your branch address name i am writing this letter to request you to kindly change the home branch of account no bank account number to new home branch with details mentioned below

bank account transfer to another branch letter format letters - Jan 29 2023

web here is the example of bank account transfer to another branch letter format your name your address city state zip code date bank name bank address city state zip code dear sir madam i am writing this letter to request the transfer of my bank account from the current branch name to the new branch name

letter format bank account transfer from one branch to another - Jul 03 2023

web aug 29 2015 here are two templates of letters for requesting the bank to transfer the existing account to another branch of the same bank letter from an individual person date from name of the account holder complete address phone number to the branch manager bank s name branch s name sub transfer of bank

letter to inform change of bank account semioffice com - Oct 26 2022

web email letter to inform that bank account number is changed dear sir i want to inform you that my company bank account number is changed my company s new a c number is mentioned below title bank name bank account or iban would you please update the bank account number in your record for all future transactions