

(ix) The standard potential of the cell in the following reaction is \_\_\_\_\_



(  $E_{\text{Cd}^{2+}}^{\circ} = -0.403\text{V}$ ,  $E_{\text{Cu}^{2+}}^{\circ} = 0.334\text{V}$  )

(a)  $-0.737\text{V}$

(b)  $0.737\text{V}$

(c)  $-0.069\text{V}$

(d)  $0.069\text{V}$

(x) The value of  $[\text{H}_3\text{O}^+]$  in  $\text{mol lit}^{-1}$  of  $0.001\text{M}$  acetic acid solution ( $K_a = 1.8 \times 10^{-5}$ ) is \_\_\_\_\_

(a)  $1.34 \times 10^{-1}$

(b)  $1.34 \times 10^{-2}$

(c)  $1.34 \times 10^{-3}$

(d)  $1.34 \times 10^{-4}$

**Q.2. Answer the following questions:**

[8]

(i) Write the product formed when alkyl halide reacts with silver nitrite.

(ii) Write the name of product formed, when acetone is treated with 2, 4-dinitrophenyl hydrazine.

(iii) Write the name of biodegradable polyamide copolymer.

(iv) Identify the molecularity of following elementary reaction:



(v) What is the action of selenium on magnesium metal?

(vi) Write the name of isomerism in the following complexes:



(vii) Write the name of the alloy used in Fischer Tropsch process in the synthesis of gasoline.

(viii) Henry's law constant for  $\text{CH}_3\text{Br}_{(g)}$  is  $0.159\text{ mol dm}^{-3}\text{ bar}^{-1}$  at  $25^\circ\text{C}$ . What is solubility of  $\text{CH}_3\text{Br}_{(g)}$  in water at same temperature and partial pressure of  $0.164\text{ bar}$ ?

### SECTION – B

**Attempt any EIGHT of the following questions:**

[16]

**Q.3.** Explain pseudo-first order reaction with suitable example.

**Q.4.** Write the consequences of Schottky defect with reasons.

**Q.5.** What is the action of following on ethyl bromide:

(i) Na in dry ether

(ii) Mg in dry ether

**Q.6.** Explain formation of peptide linkage in protein with an example.

**Q.7.** Derive an expression to calculate molar mass of non volatile solute by osmotic pressure measurement.

**Q.8.** Explain monodentate and ambidentate ligands with example.

**Q.9.** Explain the trends in the following atomic properties of group 16 elements:

(i) Atomic radii

# March 2physical Science Question Paper

**Judd E. Hollander**



## March 2physical Science Question Paper:

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Sampson Low,1911 Volumes for 1898 1968 include a directory of publishers      **The Chemical World** ,1914      *The*  
*Academy* ,1895      Science John Michels (Journalist),1888 Since Jan 1901 the official proceedings and most of the papers of  
the American Association for the Advancement of Science have been included in *Science*      *Chemical News and Journal of*  
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web in 1998 he was presented the kodansha manga award s comic of the year for his work on gambling apocalypse kaiji known to incorporate both elements of gambling and finance in his works fukumoto s most memorable titles tend to focus on the consequences of japan s economic culture

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web kodansha collected its chapters in thirteen tankōbon volumes released from september 6 1996 to october 6 1999 it was followed by the second part tobaku hakairoku kaiji gambling apocalypse kaiji was adapted by madhouse into an anime television series kaiji ultimate survivor which was broadcast from 2007 to 2008