

# Appendix 1

## Abbreviations and symbols for quantities and units

Unavoidably, some symbols are used for more than one thing. The meaning is usually clear from the context, but always check that you understand what a symbol is representing. Note that symbols representing *physical quantities* are printed in *italics*. Symbols for units are *not* in italics. The reference (in brackets) given for each entry is to chapters where the abbreviation or symbol is used.

|                     |  |                  |   |
|---------------------|--|------------------|---|
| A                   | ampere (unit of electrical current) (1, 14)                      | °C               | degrees Celsius (1)   |
| A                   | absorbance (10, 11, 28)  | $C_p$            | specific heat capacity (13)   |
| A                   | area (8, 13, Maths Toolkit MTS)                                  | $c$              | concentration of a solution (1)   |
| A                   | frequency factor (in Arrhenius equation) (9)                     | $c$              | speed of light (2, 10)  |
| A                   | Madelung constant (6)  | $\bar{c}$        | root mean square speed (of molecules) (8)                                       |
| $A_r$               | relative atomic mass (1)   | $\bar{c}$        | mean speed (of molecules) (8)   |
| Å                   | angstrom (non-SI unit used for bond lengths) (1)                 | CFSE             | crystal field stabilization energy (28)   |
| $a$                 | acceleration (1, 8)  | CI               | chemical ionization (in mass spectrometry) (12)                                 |
| $a$                 | thermodynamic activity (14, 15, 16, 17)                          | CIP              | Cahn-Ingold-Prelog (sequence rules for R/S nomenclature of chiral centres) (18) |
| $a$                 | van der Waals constant for gases (8)                             | Cbz              | carboxybenzyl group (24)  |
| $a_0$               | Bohr radius (3)  | cm               | centimetre (unit of length) (1)   |
| AAS                 | atomic absorption spectrometry (11)                              | $\text{cm}^3$    | cubic centimetre (unit of volume) (1)   |
| AES                 | atomic emission spectrometry (11)                                | $\text{cm}^{-1}$ | reciprocal centimetre (unit of wavenumber) (10, 12)                             |
| AO                  | atomic orbital (4)   | D                | debye (non-SI unit of electric dipole moment) (5, 17)                           |
| Ac                  | acetyl (2)   | D                | bond dissociation enthalpy (4, 13)  |
| Ar                  | general aryl group (2)   | $\bar{D}$        | mean bond enthalpy (13)   |
| [acac] <sup>-</sup> | pentane-2,4-dionato (acetylacetonato) (28)                       | D                | centrifugal distortion constant (in rotational spectroscopy) (10)               |
| atm                 | atmosphere (non-SI unit of pressure) (1, 8)                      | $d$              | distance (1)  |
| B                   | magnetic field strength (10)                                     | $\delta$         | doublet (in NMR spectrum) (12)  |
| B                   | rotational constant (10)   | DBU              | 1,8-diazabicyclo[5.4.0]undec-7-ene (19)   |
| $b$                 | van der Waals constant for gases (8)                             | DMAP             | 4-(dimethylamino)pyridine (19)  |
| Bn                  | benzyl (2)   | DMF              | <i>N,N</i> -dimethylformamide (2, 20)   |
| Boc                 | <i>tert</i> -butoxy carbonyl group (24)                          | DMSC             | dimethyl sulfoxide (20)   |
| Bz                  | benzoyl (2)  | Da               | dalton (atomic mass unit) (1)   |
| bar                 | bar (unit of pressure) (1, 8)                                    | dien             | 1,4,7-triazabepane (diethylenetriamine) (28)                                    |
| bipy                | 2,2'-bipyridine (28)   | $\text{dm}^3$    | cubic decimetre (unit of volume) (1)  |
| bp                  | bonding pair (5)   | E                | electrophile (19)   |
| b.p.                | normal boiling point (usually quoted in °C) (1)                  | $E$              | energy (3, 10)  |
| br                  | broad (IR absorption) (12)                                       | $E_{\text{kin}}$ | kinetic energy (1, 10)  |
| C                   | coulomb (unit of electric charge) (1, 14)                        | $E_{\text{pot}}$ | potential energy (1, 10, 17)  |
| C                   | Curie constant (in Curie's law for magnetic susceptibility) (28) |                  |   |
| $C_p$               | molar heat capacity at constant pressure (13, 14)                |                  |   |
| $C_v$               | molar heat capacity at constant volume (13, 14)                  |                  |   |

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