

## Chemistry Chapter 5 Exam

### Multiple Choice

1. Which of the following is not a transition metal?

- A. Potassium
- B. Vanadium
- C. Cobalt
- D. Manganese
- E. Copper
- F. Silver
- G. Titanium
- H. Chromium
- I. Nickel
- J. Iron

- K. The elements with atomic numbers 21-30
- L. The *d*-block elements together with the *f*-block elements, Groups 4-10, 11, 12
- M. One half of the diagonal between the main s-block elements that are located together
- N. Any property that results in the transition of an ion
- O. The way that atoms lose the regular physical and chemical properties of elements, change periodically with increasing number of their atoms
- P. The elements with atomic numbers 21-30
- Q. The *d*-block elements are metals with typical metallic properties
- R. Transition ion
- S. Properties
- T. The elements of Group 11

### Multiple Choice

2. Which type of element(s) are the most reactive?

- A. Alkali earth metals
- B. Alkali metals
- C. Halogens
- D. Transition metals

3. Which group in groups 4-10 are referred to as:

- A. Halogens
- B. Noble gases
- C. Transition metals
- D. Alkali

4. What is/are the most reactive of the transition metals?

- A. Alkali elements

# Modern Chemistry Chapter 5 Sample Tests

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