

Name _____

Date _____

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Modern Chemistry • CHAPTER 7

HOMework 7-1

(pp. 261–266)

VOCABULARY

Write *true* or *false* for each statement.

1. A **chemical formula** indicates the relative number of molecules of each kind in a chemical compound.

2. **Monoatomic ions** are ions formed from a single atom. _____
3. **Monoatomic anions** of an element are identified by the element's name. _____
4. A **binary compound** is made of two or more different elements. _____
5. The **Stock system** is a system for naming chemical elements and their ions. _____

SKILL BUILDER

Explain what each chemical formula tells you about the composition of the compound. The first one is done for you.



The subscript 3 refers to the number of oxygen atoms. There are one atom of sodium and one atom of nitrogen for every three atoms of oxygen.



STANDARDIZED TEST PREP

Circle the letter of the best answer.

1. Which statement is true?
 - a. All main-group elements are able to form ions.
 - b. The total numbers of positive charges and negative charges must be equal in a binary compound.
 - c. The nonmetals of groups 15, 16, and 17 lose electrons to form cations.
 - d. Elements in the *d* block form ions of only one charge.
2. Why do many main-group elements lose or gain electrons to form ions?
 - a. The elements are unstable and therefore lose electrons.
 - b. Their tendency is to maximize potential energy.
 - c. The elements form covalent bonds, sharing their electrons with other atoms.
 - d. Their tendency is to form a complete outermost octet in a noble-gas configuration.

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Molecular Mechanics Across Chemistry Anthony K. Rappé, Carla J. Casewit, 1997-05-07 The remarkable breadth of modern molecular mechanics is covered in this textbook developed for an undergraduate or first time course on molecular mechanics The book uses a case study approach designed to give readers exposure to the relevance and utility of molecular mechanics as well as the opportunity to study a particular problem and its solution in depth The remarkable breadth of modern molecular mechanics is covered in this textbook developed for an undergraduate or first time course on molecular mechanics The book uses a case study approach designed to give readers exposure to the relevance and utility of molecular mechanics as well as the opportunity to study a particular problem and its solution in depth **Books in Print**

Supplement, 2002 **Fundamentals of Chemistry: A Modern Introduction (1966)** Frank Brescia, 2012-12-02

Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro s Law and Graham s Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry *Children's Books in Print* R R Bowker Publishing, Bowker, 1999-12 **General Chemistry** Ralph H. Petrucci, 1989 *Chemistry in the Modern World* Frank L. Wiseman, 1985 **Resources in Education**, 1968

Proceedings of the 2024 9th International Conference on Modern Management, Education and Social Sciences (MMET 2024) Chunyi Lin, Fong Peng Chew, Intakhab Alam Khan, Catalin Popescu, 2024-12-07 This is an open access book As a leading role in the global megatrend of scientific innovation China has been creating a more and more open environment for scientific innovation increasing the depth and breadth of academic cooperation and building a community of innovation that benefits all Such endeavors are making new contributions to the globalization and creating a community of shared future To adapt to this changing world and China s fast development in the new era 2024 9th International Conference on Modern Management Education and Social Sciences MMET 2024 to be held in September 20 22 2024 This conference takes bringing together global wisdom in scientific innovation to promote high quality development as the theme and focuses on cutting

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Soviet Electrochemistry ,1975 **Glencoe Earth Science** Ralph M. Feather,1999 Earth science is the study of Earth and space It is the study of such things as the transfer of energy in Earth s atmosphere the evolution of landforms patterns of change that cause weather the scale and structure of stars and the interactions that occur among the water atmosphere and land Earth science in this book is divided into four specific areas of study geology meteorology astronomy and oceanography p 8 9

Otto E. Miller, Plaintiff-Respondent, Against Fred W. Smythe, Defendant-Appellant , *Chemistry* Martin Stuart Silberberg,1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives

Linear Algebra David Poole,2005 David Poole s innovative book emphasizes vectors and geometric intuition from the start and better prepares students to make the transition from the computational aspects of the course to the theoretical Poole covers vectors and vector geometry first to enable students to visualize the mathematics while they are doing matrix operations With a concrete understanding of vector geometry students are able to visualize and understand the meaning of the calculations that they will encounter By seeing the mathematics and understanding the underlying geometry students develop mathematical maturity and can think abstractly when they reach vector spaces Throughout the text Poole s direct conversational writing style connects with students and an abundant selection of applications from a broad range of disciplines clearly demonstrates the relevance of linear algebra

Secondary Education Journal ,1985 School ,1906 *Forthcoming Books* Rose Arny,2003 **Choice** ,1997 Secondary Teaching Methods Kenneth T. Henson,1981 *American Scientist* ,1942

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