

Molar Mass worksheet

Determine the molar mass for the following elements:

1 mole of an element = the atomic mass of the element in grams

Ca

Se

K

Fe

As

O

Determine the molar mass for the following molecular compounds:

1 mole of a compound = the masses of each element times the number of atoms of each element combined

NaCl

H_2O

O_2

Fe_2O_3

Ba_2O

Cl_2

$\text{Al}_2(\text{SO}_4)_3$

$\text{Ca}(\text{NO}_3)_2$

MgCl_2

Determine the mass of the following samples:

2 moles Co

5 moles Ba

3 moles H_2O

Determine the number of moles in the following samples:

54.2 g Ba

200.2 g MgO

519.2 g FeCl_2

Determine how many atoms/molecules in the following samples:

1.32 moles Si

7.29 mol CO_2

712.4 g NaCl

Molar Mass Of Elements Answer Key

L Reisser



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