

## Molality Practice

1. Calculate the molarity of each of the following solutions:

a. 12.4 g KCl in 289.2 mL solution

$$M = \frac{n}{V} \quad 12.4 \text{ g KCl} \times \frac{1 \text{ mole KCl}}{74.6 \text{ g KCl}} = 0.166 \text{ mole KCl}$$

$$289.2 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 0.2892 \text{ L}$$

$$\frac{0.166 \text{ mole}}{0.2892 \text{ L}} = 0.57 \text{ M}$$

## Practice Problem

b. 16.4 g  $\text{CaCl}_2$  in 0.614 L solution

$$M = \frac{n}{V} \quad 16.4 \text{ g CaCl}_2 \times \frac{1 \text{ mole CaCl}_2}{111.1 \text{ g CaCl}_2} = 0.148 \text{ mole CaCl}_2$$

$$M = \frac{0.148 \text{ mole}}{0.614 \text{ L}} = 0.24 \text{ M}$$

# Molarity And Molality Notes And Practice Answers

**G Thomas**



## **Molarity And Molality Notes And Practice Answers:**

## Unveiling the Power of Verbal Beauty: An Psychological Sojourn through **Molarity And Molality Notes And Practice Answers**

In a global inundated with screens and the cacophony of instant communication, the profound energy and psychological resonance of verbal beauty usually fade into obscurity, eclipsed by the continuous onslaught of sound and distractions. Yet, nestled within the musical pages of **Molarity And Molality Notes And Practice Answers**, a interesting work of literary brilliance that pulses with organic thoughts, lies an unforgettable journey waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this exciting opus manuals readers on an emotional odyssey, gently revealing the latent possible and profound impact stuck within the complex web of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book is key subjects, dissect its interesting publishing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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