

## Concentration and Dilution Activity

You will be using a PHET simulation to explore the concept of molarity, and practice molarity and dilution calculations.

**Purpose of this activity:**

- Skills practiced: quantitative analysis of solutions
- Content reviewed: molarity and dilutions

**Tasks:**

- Complete the following activity.

**Criteria for Success:**

- Follow the directions below and complete the questions and calculations. Make sure to show your calculations and work for credit. Grade is based on completion, not accuracy.

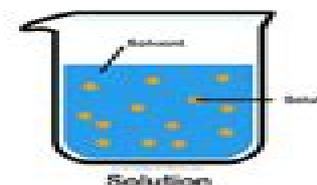
**Simulation link:**

[Concentration Simulation Link PHET Concentration](#)

**Background information:**

Much of chemistry takes place in **solution**. A solution has three parts:

- **Solute** – substance dissolved in another substance
- **Solvent** – the substance (usually liquid) in which the solute(s) is/are dissolved
- **Solute + Solvent = Solution**



An **aqueous** solution is a solution in which the solvent is water.

The most common unit of solution concentration is **molarity**. The definition of molarity is given below.

$$\text{Molarity (M)} = \frac{\text{moles of solute (mol)}}{\text{volume of solution (L)}}$$

Our goal is to use a simulation to explore both **qualitative** and **quantitative** aspects of solutions. Follow the **instructions** below.

Open the simulation link. At the right of the screen, drag the Concentration sensor and place it in the initial solution. At the top right corner, change your solute to cobalt (II) nitrate. Shake the container 2 – 3 times to get a concentration reading of at least 0.300 mol/L.

First, let's examine some **qualitative properties** of solutions. We will examine how different actions affect the concentration of the solution.

1. Record your initial concentration.

It has a concentration of .300 mol/l , the color of the liquid is pink

2. Now we will add more solute. Shake some more cobalt (II) nitrate into the solution. Record your new concentration.

It has a concentration of 0.960 mol/ml, the liquid always gets a little darker.

# Molarity Concepts Assignment Packet Answer Key

**Ralph Thomas Myers**



## **Molarity Concepts Assignment Packet Answer Key:**

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