

# Molarity Worksheet

1. Calculate the mass of  $\text{CuSO}_4 \cdot 6\text{H}_2\text{O}$  required to prepare 200.0 mL of 0.300 M solution.
2. Calculate the mass of  $\text{CoCl}_2 \cdot 8\text{H}_2\text{O}$  required to prepare 300.0 mL of a 0.520 M solution.
3. 150.0 g of NaCl is dissolved in 250.0 mL of water. Calculate the molarity.
4. 25.2 g of  $\text{CuSO}_4 \cdot 6\text{H}_2\text{O}$  is dissolved in 28.0 mL of water. Calculate the molarity.
5. Calculate the mass of NaCl required to prepare 565.0 mL of a 0.450 M solution.
6. Calculate the volume of 0.250 M NaCl solution required to contain 0.0300 g of NaCl.
7. Calculate the volume of 0.500 M NaCl solution required to contain 0.52 g of NaCl.
8. Calculate the mass of NaCl required to prepare 360.0 mL of a 0.35 M solution.
9. 55.6 g of NaCl is dissolved in 562 mL of water. Calculate the molarity.
10. 78.9 g of  $\text{CuSO}_4 \cdot 8\text{H}_2\text{O}$  is dissolved in 500.0 mL of water. Calculate the molarity.

# Molarity Practice Tesccc Unit Lesson 2

**H Kauffman**



## **Molarity Practice Tesccc Unit Lesson 2:**

## Decoding **Molarity Practice Tesccc Unit Lesson 2**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Molarity Practice Tesccc Unit Lesson 2**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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