

Name: \_\_\_\_\_  
 Period: \_\_\_\_\_

Fall Semester Review – IPC

Fill in the table on the right from your periodic table.

Element	Chemical Symbol	Atomic #	Valence Electrons	Oxidation #	# of Protons	Atomic Mass	Metal or Nonmetal	Group or Luster Electrons	# of Full Electron Levels
Oxygen									
Argon									
Calcium									
Nitrogen									
Potassium									
Sulfur									
Helium									

Find the molar mass of the following compounds

Calcium oxide (CaO) \_\_\_\_\_  
 Water \_\_\_\_\_

Classify (ionic, covalent or polyatomic) and name the following compounds.

- CO<sub>2</sub> \_\_\_\_\_
- N<sub>2</sub>O \_\_\_\_\_
- NaClO<sub>2</sub> \_\_\_\_\_

Draw the Lewis Dot Diagrams for

- Carbon
- Sodium

Make balanced ionic compounds from:

- Li + O \_\_\_\_\_
- Be + NO<sub>3</sub><sup>-</sup> \_\_\_\_\_
- Na + Ar \_\_\_\_\_
- Ca + O \_\_\_\_\_

With dot diagrams draw the covalent bond of O<sub>2</sub>

With dot diagrams the covalent bond of CH<sub>4</sub>

Matching:

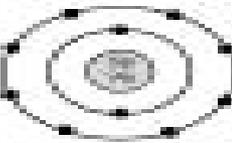
- Mixture
- Compound
- Element
- Matter
- Heterogeneous
- Homogeneous

- \_\_\_\_\_ A substance made up of two or more atoms that must be separated by chemical means.
- \_\_\_\_\_ Something made up of 2 or more different atoms that can be separated by physical means.
- \_\_\_\_\_ A substance that is pure and made up of only one type of atom.
- \_\_\_\_\_ Anything that has mass and takes up space.
- \_\_\_\_\_ A mixture that is the same throughout.
- \_\_\_\_\_ A mixture that is different throughout.

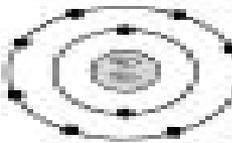
1. Valence electrons are the outermost electrons of an atom that are involved in chemical bonding? True/False

2. Lithium has \_\_\_\_\_ valence electrons, will \_\_\_\_\_ (lose or gain) \_\_\_\_\_ electrons and become \_\_\_\_\_ (positive or negative). This is why lithium's oxidation number is \_\_\_\_\_.

3. A metal and non-metal will form a \_\_\_\_\_ compound, two non-metals form a \_\_\_\_\_ compound.



Atom A



Atom B

Which one is the ion?  
 Give the ion notation for the ion.

Matching:

- Ion
- Element
- Isotope
- Oil

- \_\_\_\_\_ If you change the number of protons, you change the \_\_\_\_\_.
- \_\_\_\_\_ If you change the number of neutrons, you change the \_\_\_\_\_.
- \_\_\_\_\_ If you change the number of electrons, you change the \_\_\_\_\_.
- \_\_\_\_\_ Every 3,000 miles you should change a car's \_\_\_\_\_.

Matching:

- Unit of Measurement of Mass
- Meter
- Conductor
- Leaf
- Claw

- \_\_\_\_\_ Unit of length; equal to 3.3 ft.
- \_\_\_\_\_ Mass is neither destroyed or created in chemical reactions.
- \_\_\_\_\_ Unit of area; about 1 dollar bill.
- \_\_\_\_\_ 1,000's of mass; width of pinky finger.
- \_\_\_\_\_ Unit of volume; just bigger than a quart.

Matching:

- Solid
- Liquid
- Gas
- Oil

- \_\_\_\_\_ Molecules that are tightly packed and retain their shape and size.
- \_\_\_\_\_ Molecules that bounce off of each other, can be compressed, and take the shape of their container.
- \_\_\_\_\_ Molecules that can move (slide) over each other and have a definite size (volume), but not shape and can not be compressed.

# Fall Semester Review Ipc Answers 2014

**E Durkheim**



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