

## Assessment 3

**Chapter Test B****Chapter: Chemical Equations and Reactions**

**Part 1** In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. The production of a slightly soluble solid compound by a double-displacement reaction is called the formation of a
- gas.
  - precipitate.
  - combustion reaction.
  - halogen.
- \_\_\_\_\_ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
  - subscripts.
  - formula of the products.
  - number of products.
- \_\_\_\_\_ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- solid.
  - dissolved in water.
  - an acid.
  - molten.
- \_\_\_\_\_ 4. The tendency for a replacement reaction to occur depends on the
- activity between any two elements in the activity series themselves.
  - temperature difference.
  - valence electrons and mass of.
  - activity between any two elements in the activity series themselves.
- \_\_\_\_\_ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
  - show the number of grams of each substance that react.
  - are the molar masses of the substances.
  - show the valence electrons for each atom.
- \_\_\_\_\_ 6. If metal X is more reactive than metal Y in the activity series, then
- X will replace ions of Y in solution.
  - Y will replace ions of X in solution.
  - Y will form oxides only indirectly.
  - X will react with cold water.

# Guide Key Chemistry

**D Keegan**



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**Organic Chemistry Study Guide** Robert J. Ouellette, J. David Rawn, 2014-10-13 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place [The Key](#) Rao Gautam, 2003 [The Key](#) Simonne Longerich, 1998 **Organic Chemistry Study Guide and Solutions** Marc Loudon, Jim Parise, 2015-07-01 Parise and Loudon's Study Guide and Solutions Manual offers the following learning aids Links that provide hints for study approaches to problem solving and additional explanations of challenging topics Further Explorations that provide additional depth on key topics Reaction summaries that delve into key mechanisms and stereochemistry Solutions to all the textbook problems Rather than providing just the answer many of the solutions provide detailed explanations of how the problem should be approached [The Key](#), 2004 [Catalog of Copyright Entries. Third Series](#) Library of Congress. Copyright Office, 1978 *Jackson's compendium, a guide-book for the teaching of handwriting* John Jackson (F.E.I.S.), 1895 [The Key](#) Simonne Longerich, 2002 [The Key Student Study Guide](#), 2023 **A Laboratory Guide for Students in Physical Sciences** Heinrich Schapper, 1908 **A Self-study Guide to the Principles of Organic Chemistry** Jiben Roy, 2013 A Self Study Guide to the Principles of Organic Chemistry Key Concepts Reaction Mechanisms and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily as well as build a strong foundation for future study Starting with the definition of atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix [The Key](#) Mun Prasad, 2004 **Manual of Psychiatry** Joseph Rogues de Fursac, 1908 [The Key](#), 2004 **The Key** Gautam

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